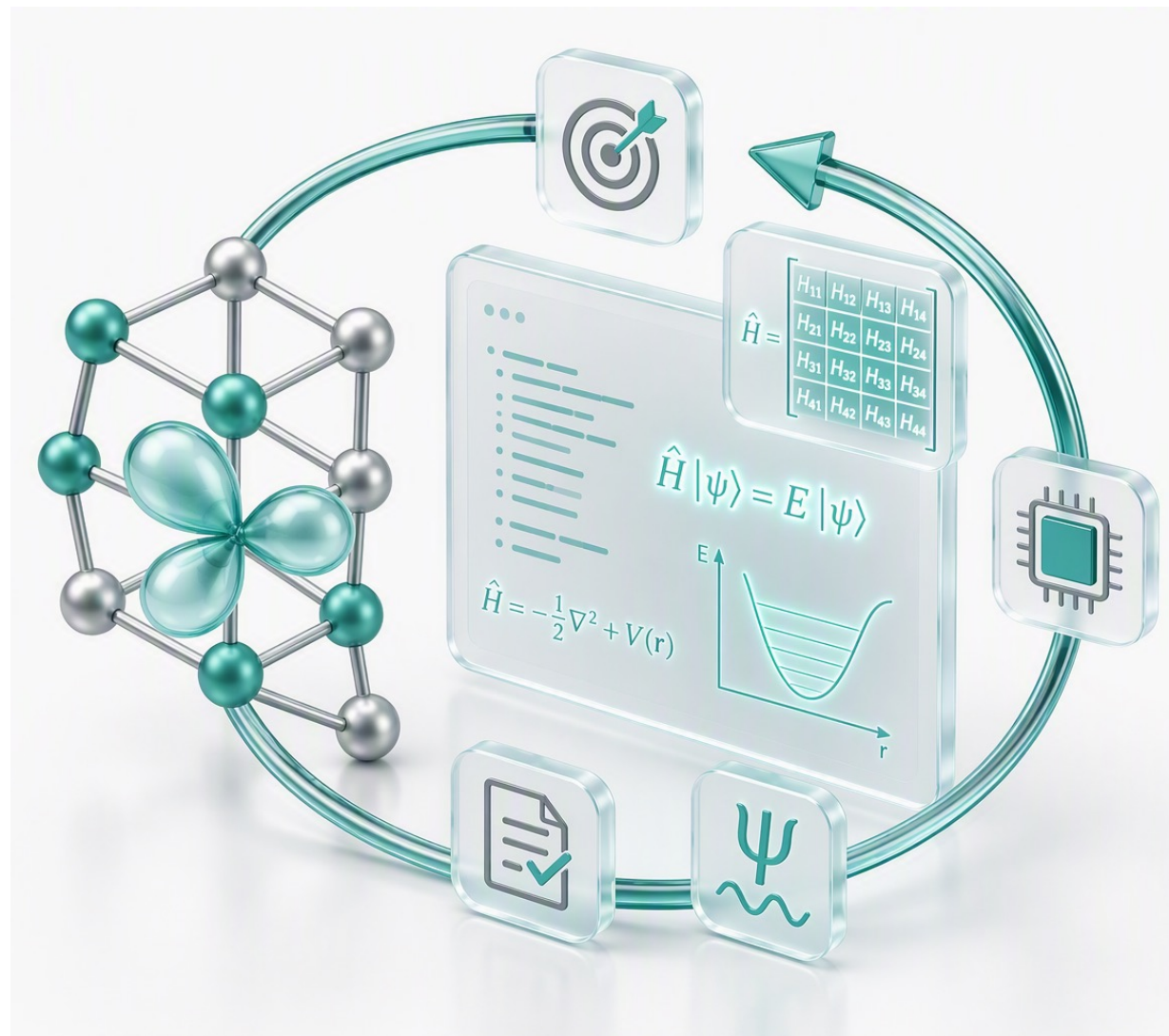




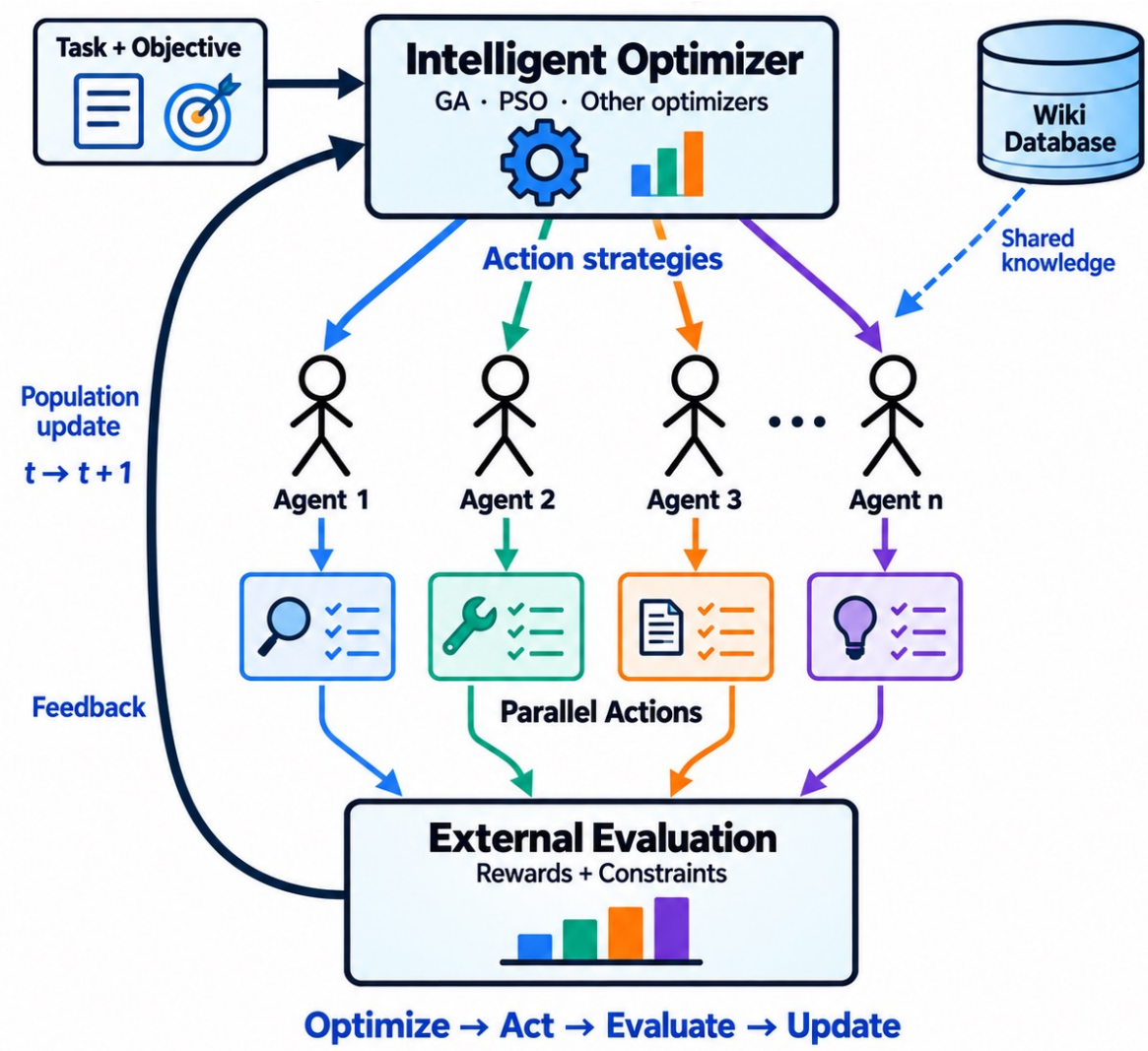
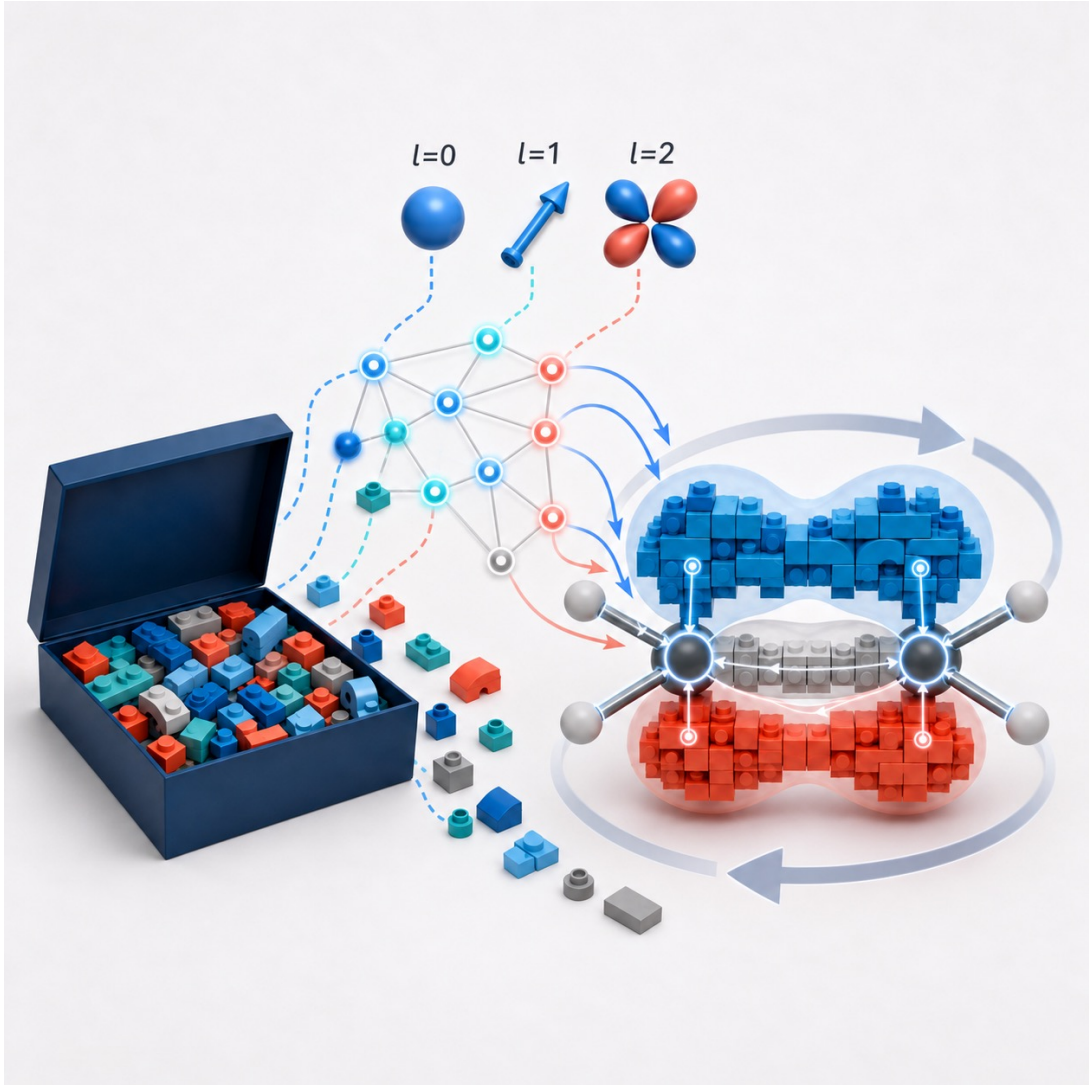
Yuan Jiao

School of Advanced Interdisciplinary Sciences, SAIS
University of Chinese Academy of Sciences

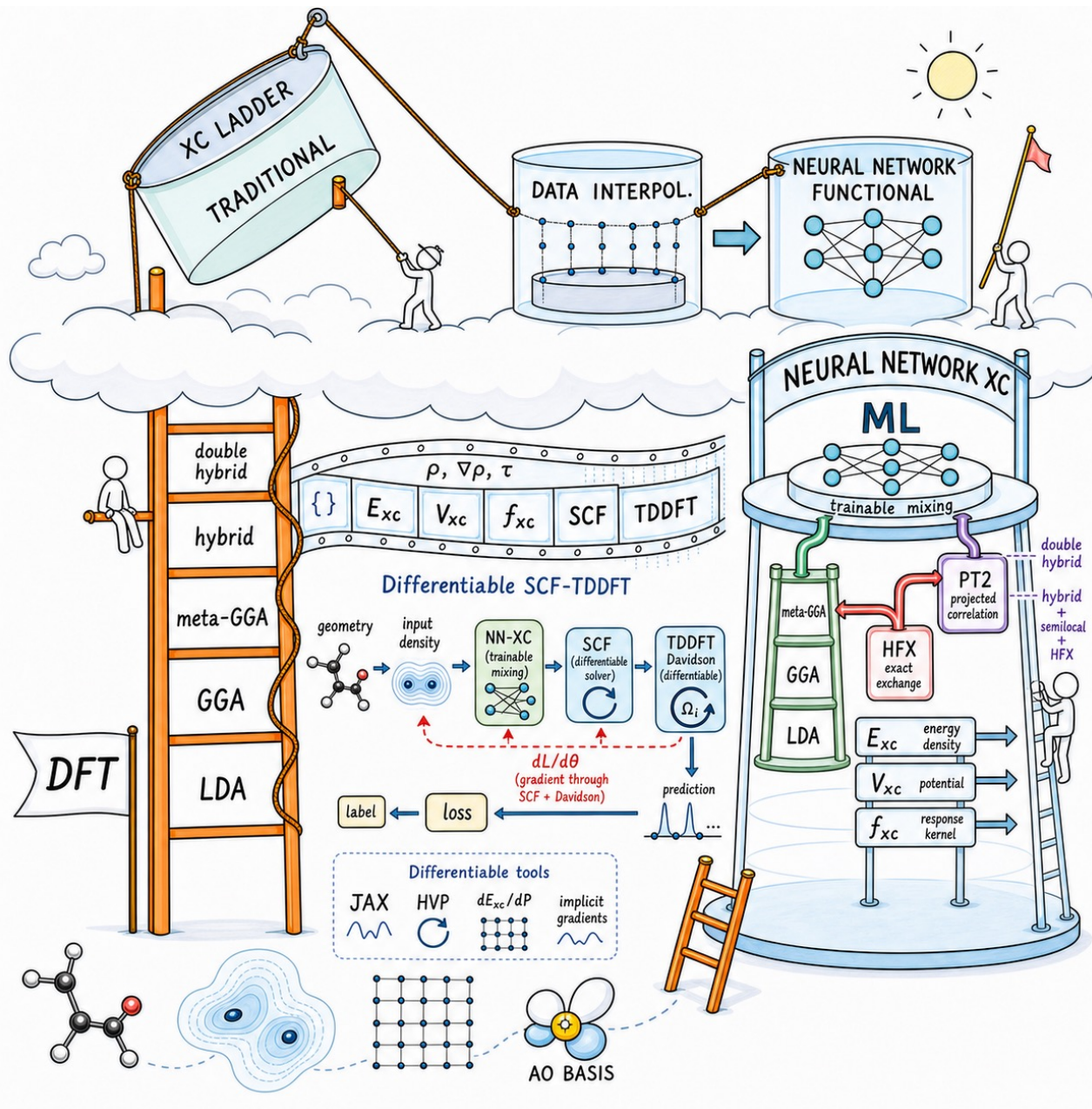


■ Environment-Adaptive Atomic Orbital Basis Function

■ Agent Auto-Research for Molecular Designing



Background-From GradTDDFT to GradSCF



• One-Electron Integral

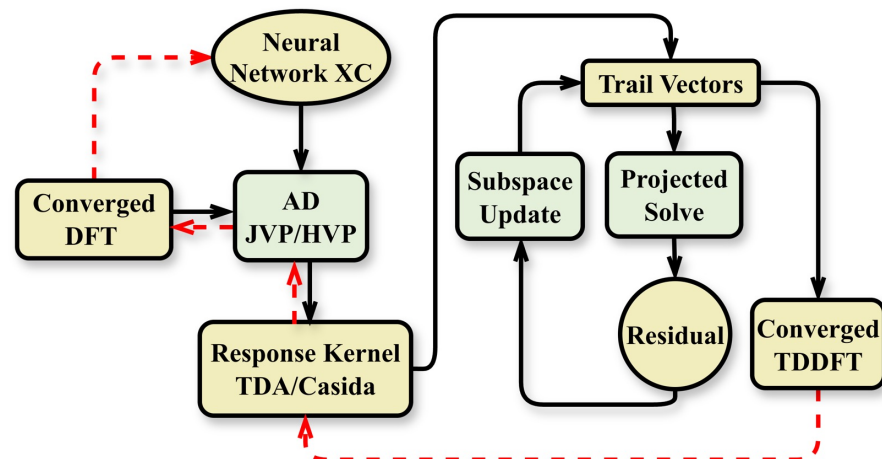
$$S_{\mu\nu} = \int \chi_{\mu}(r) \chi_{\nu}(r) dr \quad T_{\mu\nu} = \int \chi_{\mu}(r) \nabla^2 \chi_{\nu}(r) dr$$

$$h_{\mu\nu} = \int \chi_{\mu}(r) \hat{h} \chi_{\nu}(r) dr \quad v_{\mu\nu} = \int \chi_{\mu}(r) \frac{1}{r} \chi_{\nu}(r) dr$$

• Two-Electron Integral

$$(\mu\nu|\lambda\sigma) = \int \frac{\chi_{\mu}^*(r_1) \chi_{\nu}(r_1) \chi_{\lambda}^*(r_2) \chi_{\sigma}(r_2)}{r_{12}} dr_1 dr_2.$$

• HF/DFT and After Single Particle Orbital



■ Atomic Orbital Basis Function

➤ Slater-Type Orbitals(STOs)

$$\phi_{\zeta n \ell m}^{\text{STO}}(\mathbf{r}) \propto r^{n-1} e^{-\zeta r} Y_{\ell m}(\hat{\mathbf{r}})$$

$$\phi_{1s}^{\text{STO}}(r) \propto e^{-\zeta r} \quad \left. \frac{d\phi_{1s}^{\text{STO}}(r)}{dr} \right|_{r=0} = -\zeta$$

- atom-centered
- Natural nuclear cusp and exponential tail

➤ Numerical Atomic Orbitals(NAOs)

- Radial functions tabulated numerically
- Compact and flexible atom-centered basis

➤ Gaussian-Type Orbitals(GTOs)

$$\phi_{\alpha n \ell m}^{\text{GTO}}(r) \propto r^{2(n-1)-\ell} e^{-\alpha r^2} Y_{\ell m}(\hat{\mathbf{r}})$$

$$\phi_{1s}^{\text{GTO}}(r) \propto e^{-\alpha r^2} \quad \left. \frac{d\phi_{1s}^{\text{GTO}}(r)}{dr} \right|_{r=0} = 0$$

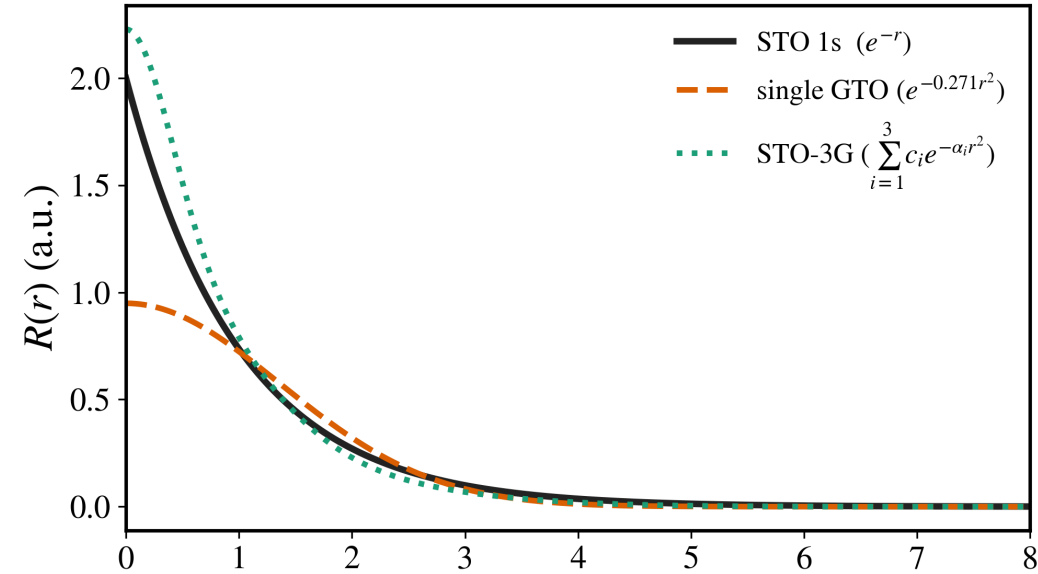
- atom-centered
- Standard choice for molecular calculations
- Efficient evaluation of multicenter integrals

➤ Plane Waves(PWs)

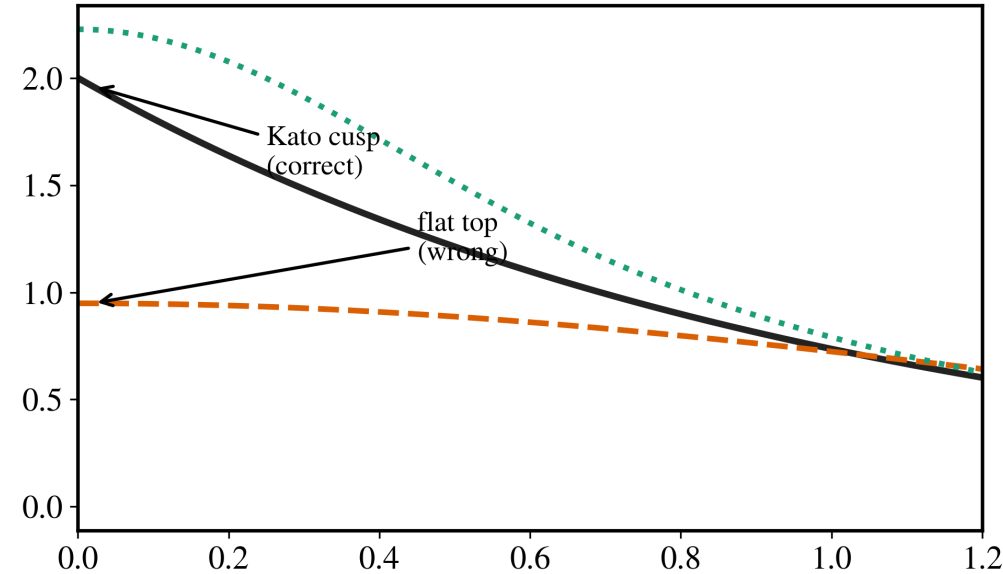
- Delocalized and orthonormal
- Naturally adapted to periodic systems
- Systematically converged with an energy cutoff

Atomic Orbital Basis Function

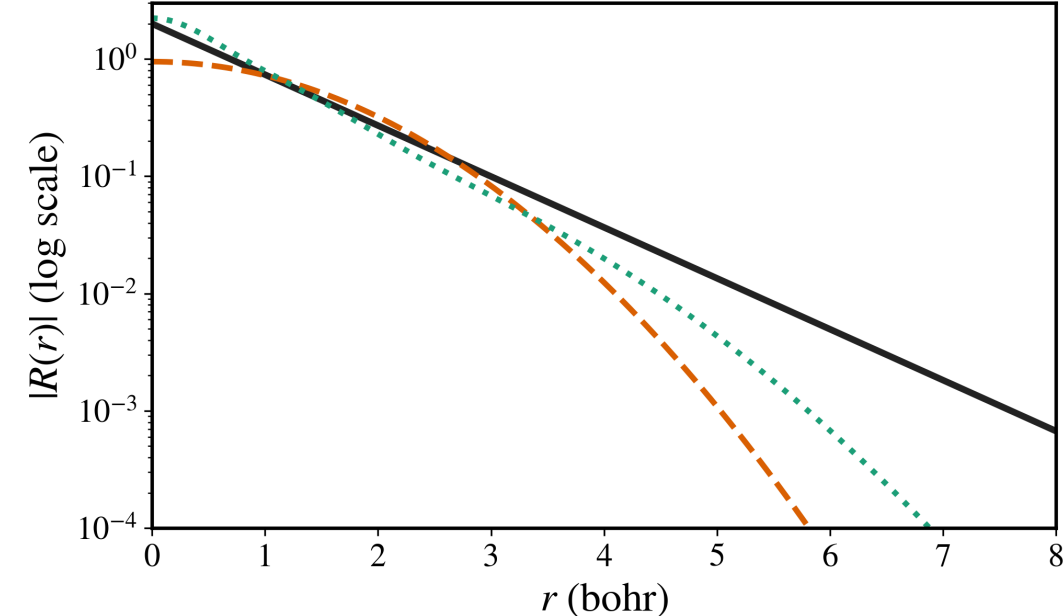
(a) 1s radial functions



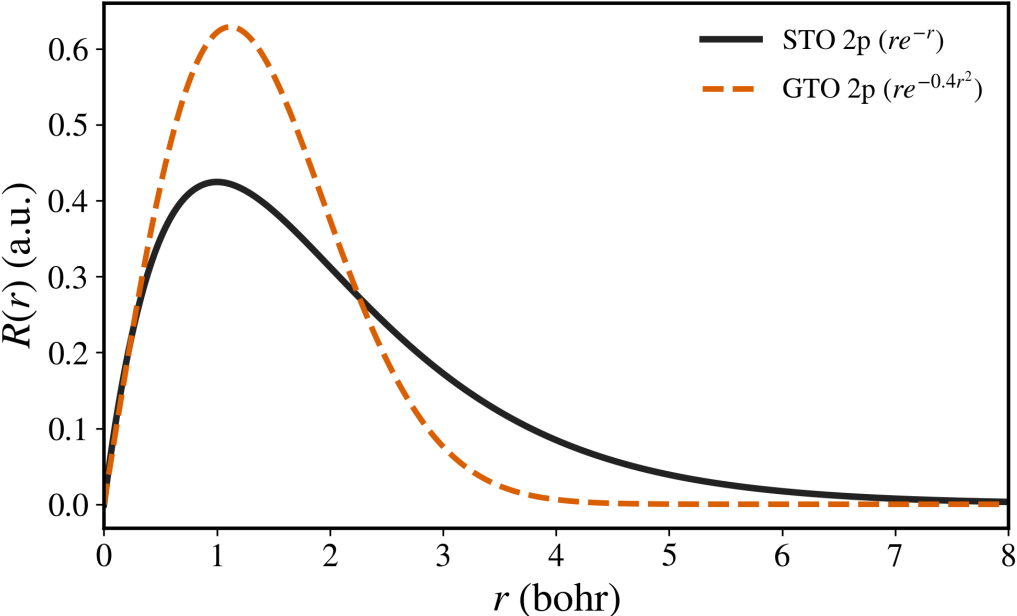
(b) cusp region: STO has $R'(0) \neq 0$, GTOs have $R'(0) = 0$



(c) tail: Gaussian $e^{-\alpha r^2}$ decays too fast



(d) 2p radial functions



■ Atomic Orbital Basis Function

➤ Basis Set Family

- **Valence flexibility:** use double-, triple-, or higher- ζ descriptions to give each valence orbital several radial shapes
 - **Polarization functions:** add higher angular momenta to describe anisotropic density, bonding, and angular correlation
 - **Diffuse functions:** add small exponents to describe long-range density (anions, Rydberg states, noncovalent interactions, multipoles, (hyper)polarizabilities, etc)
 - **Tight functions:** add large exponents to describe core region (core correlation, nuclear magnetic shielding, spin-spin coupling constants, etc)
-

6-> Six primitive functions for core orbital

3-> three primitive functions for inner orbital

1-> one primitive functions for valence orbital

++->diffusion primitive function

******->polarized primitive function

6-311++G**

■ Atomic Orbital Basis Function

➤ Basis Set Family

cc-pVDZ: 3s2p1d

s	s	s
p	p	
d		

aug-cc-pVDZ: 4s3p2d

s	s	s	s
p	p	p	
d	d		

d-aug-cc-pVDZ: 5s4p3d

s	s	s	s	s
p	p	p	p	
d	d	d		

cc-pVTZ: 4s3p2d1f

s	s	s	s
p	p	p	
d	d		
f			

aug-cc-pVTZ: 5s4p3d2f

s	s	s	s	s
p	p	p	p	
d	d	d		
f	f			

d-aug-cc-pVTZ: 6s5p4d3f

s	s	s	s	s	s
p	p	p	p	p	
d	d	d	d		
f	f	f			

cc-pVQZ: 5s4p3d2f1g

s	s	s	s	s
p	p	p	p	
d	d	d		
f	f			
g				

aug-cc-pVQZ: 6s5p4d3f2g

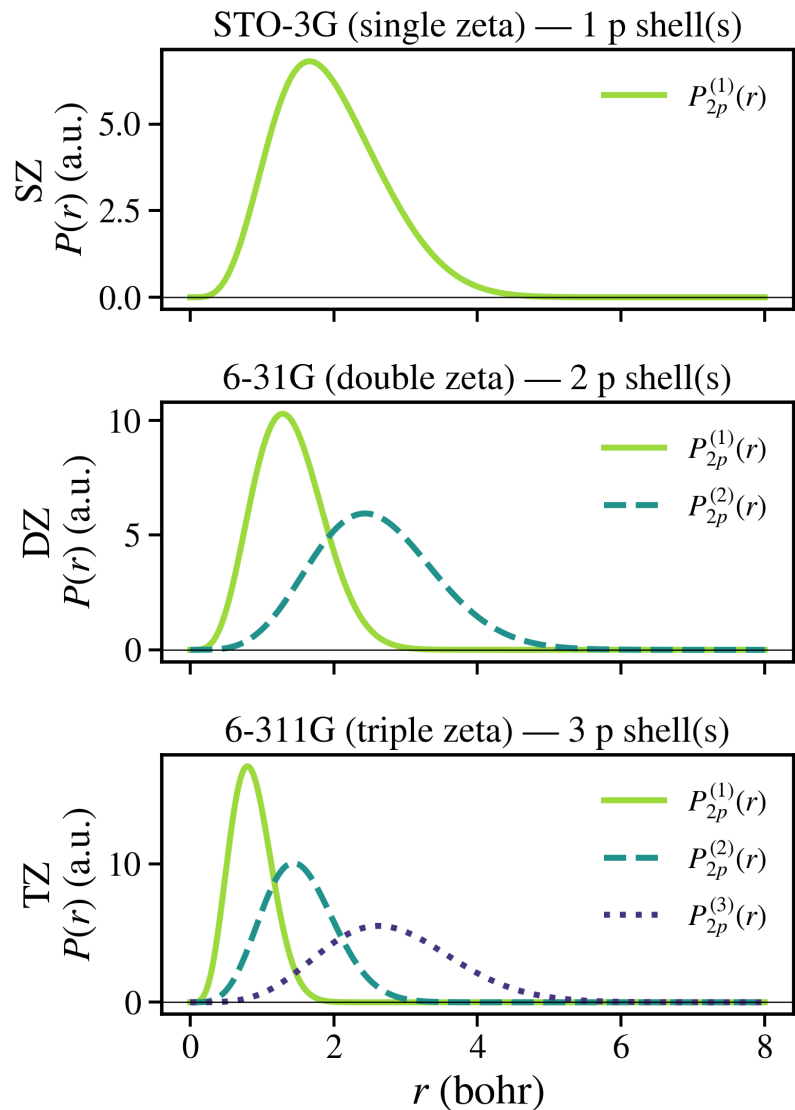
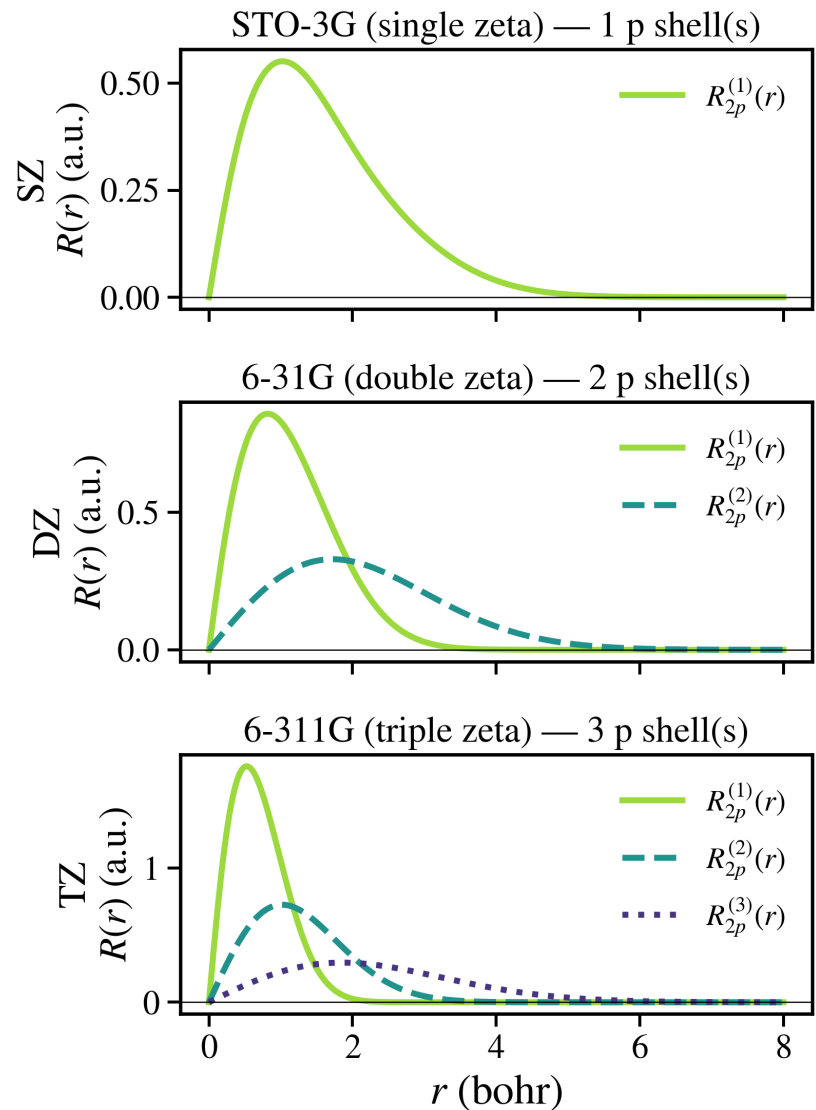
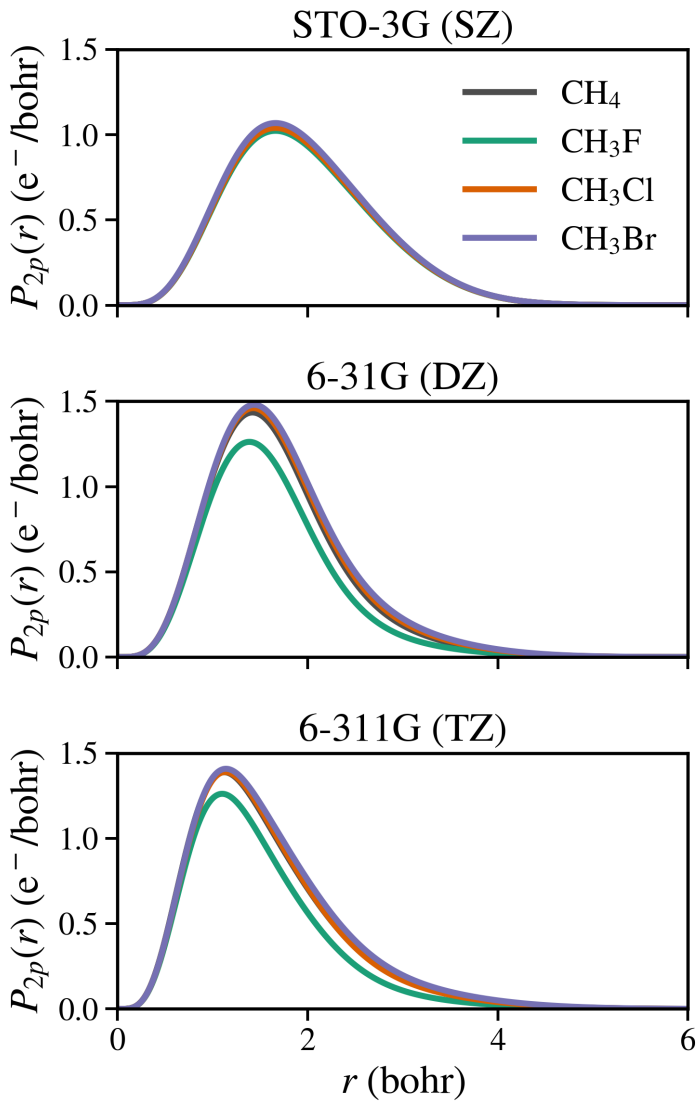
s	s	s	s	s	s
p	p	p	p	p	
d	d	d	d		
f	f	f			
g	g				

d-aug-cc-pVQZ: 7s6p5d4f3g

s	s	s	s	s	s	s
p	p	p	p	p	p	
d	d	d	d	d		
f	f	f	f			
g	g	g				

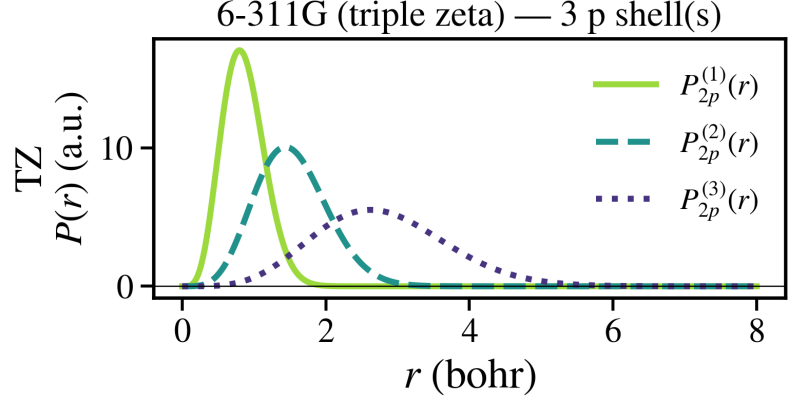
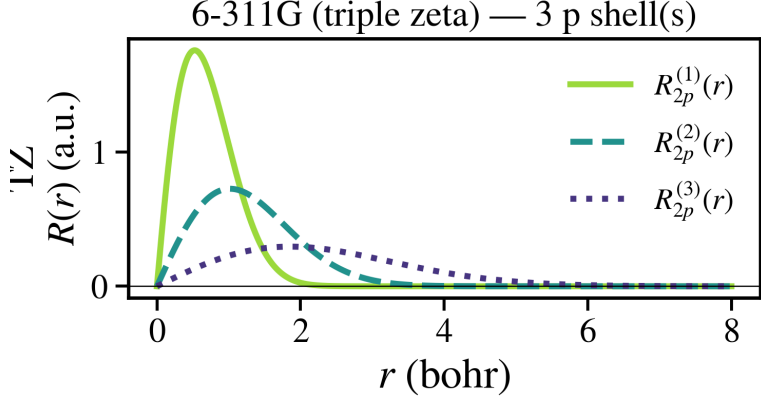
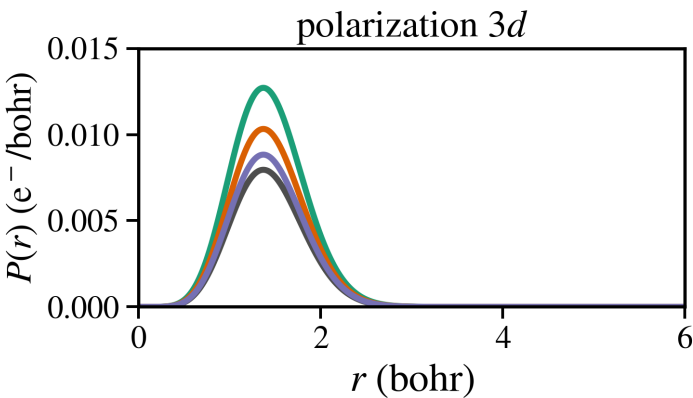
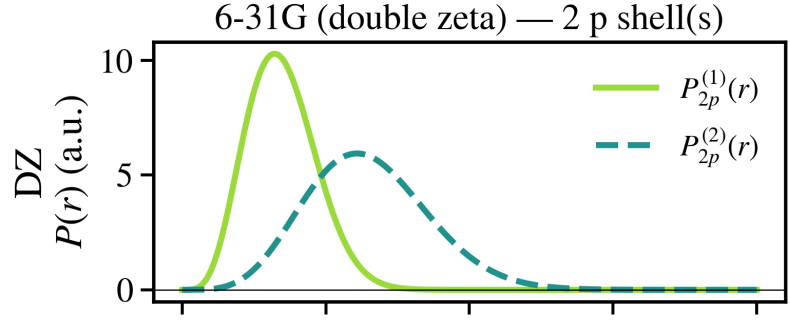
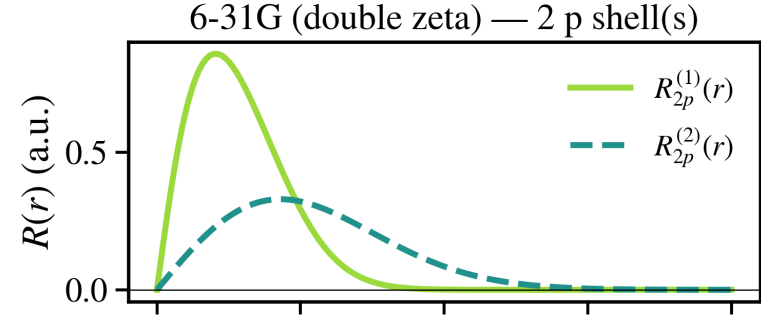
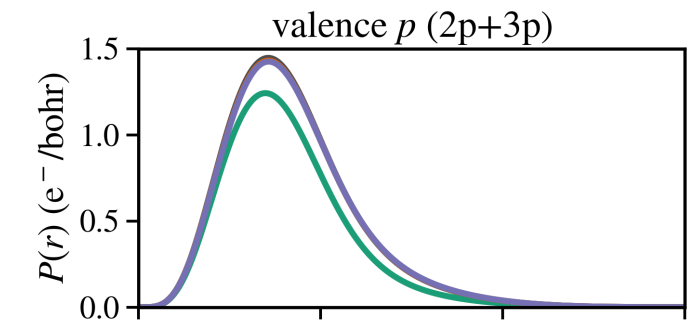
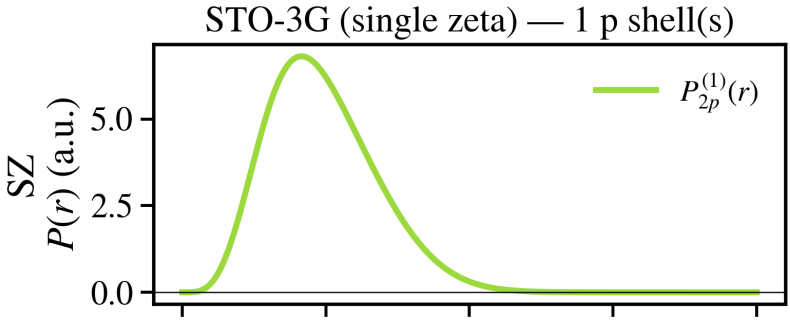
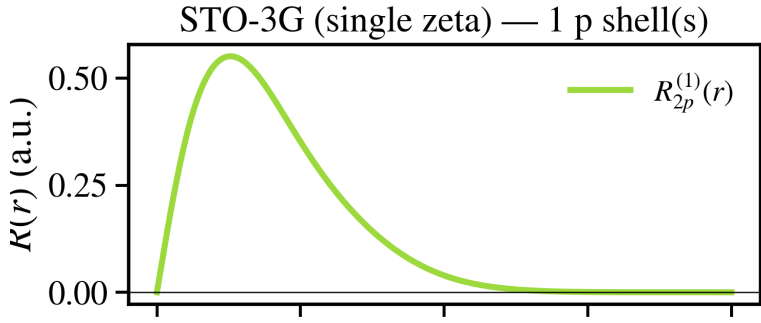
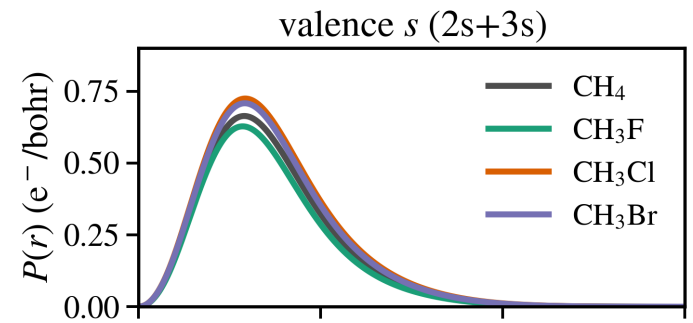
Atomic Orbital Basis Function

Contracted Multi-Zeta Basis Set



Atomic Orbital Basis Function

➤ Contracted Multi-Zeta Basis Set



■ Chemical Environment-Adaptive Basis

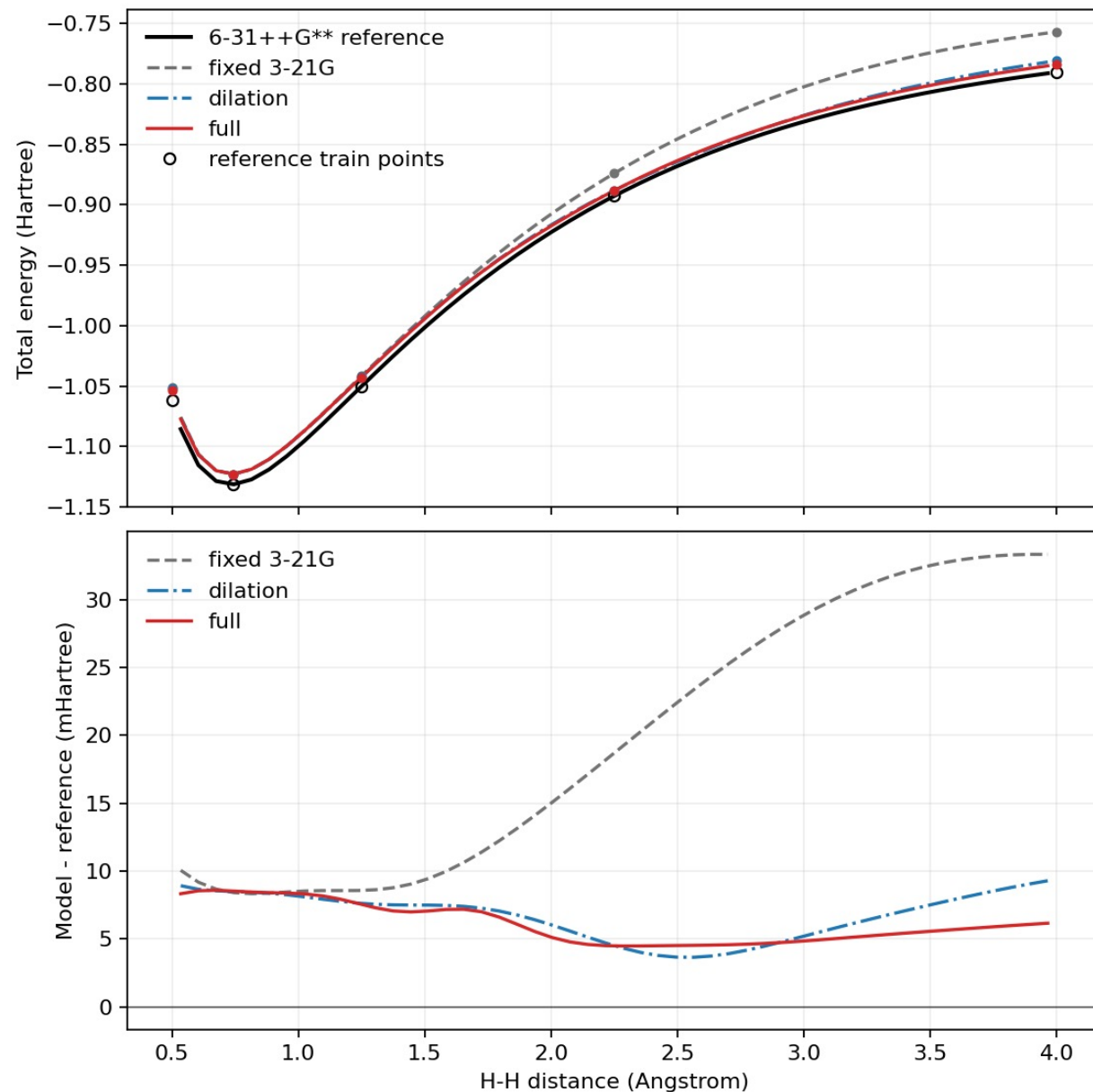
➤ 1: Neural Network Atom Basis Designing and Training

$$\phi_{c,\rho}^0(r) = \sum_p d_{cp} N_s(\rho_c^2 \alpha_{cp}) \exp(-\rho_c^2 \alpha_{cp} r^2)$$

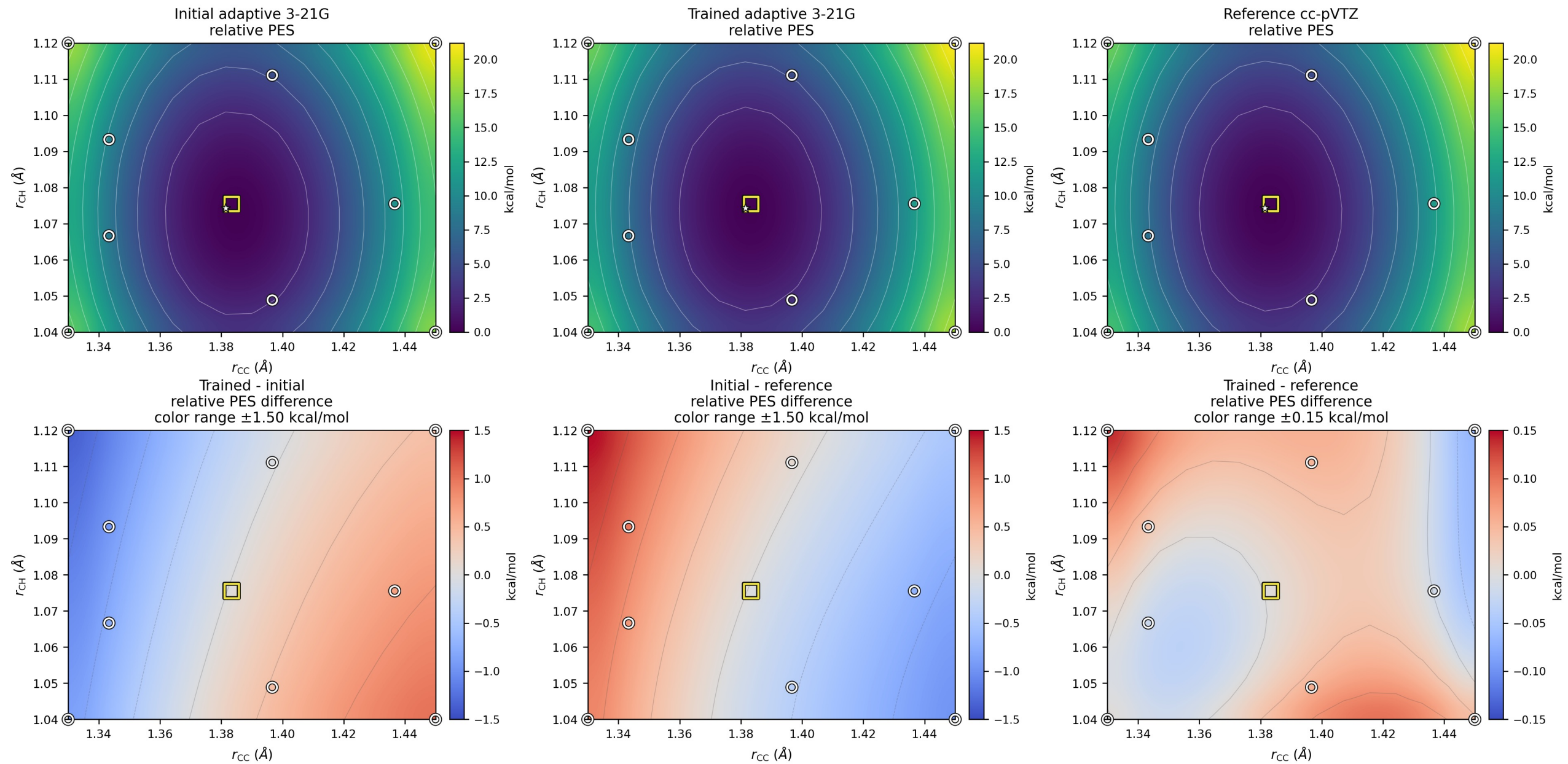
$$\rho_{Ac} = \exp[(\ln 2) \tanh(\eta_{Ac})], \quad \rho_{Ac} \in (0.5, 2)$$

$$\eta_{Ac} \rightarrow \{\phi_{c,\rho}^0(r)\} \rightarrow \{S, h, (\mu\nu|\lambda\sigma)\} \rightarrow SCF \rightarrow observables$$

atom	shell	primitive	basis coefficient / bohr ⁻²	model
C	core S(3)	3	172.256, 25.9109, 5.53335	fixed
C	valence S(2)	2	3.66498, 0.770545	$\times(\lambda_C^2)$
C	valence S(1)	1	0.195857	$\times(\lambda_C^2)$
C	valence P(2)	2	3.66498, 0.770545	$\times(\lambda_C^2)$
C	valence P(1)	1	0.195857	$\times(\lambda_C^2)$
H	S(2)	2	5.447178, 0.824547	$\times(\lambda_C^2)$
H	S(1)	1	0.183192	$\times(\lambda_C^2)$



■ Chemical Environment-Adaptive Basis

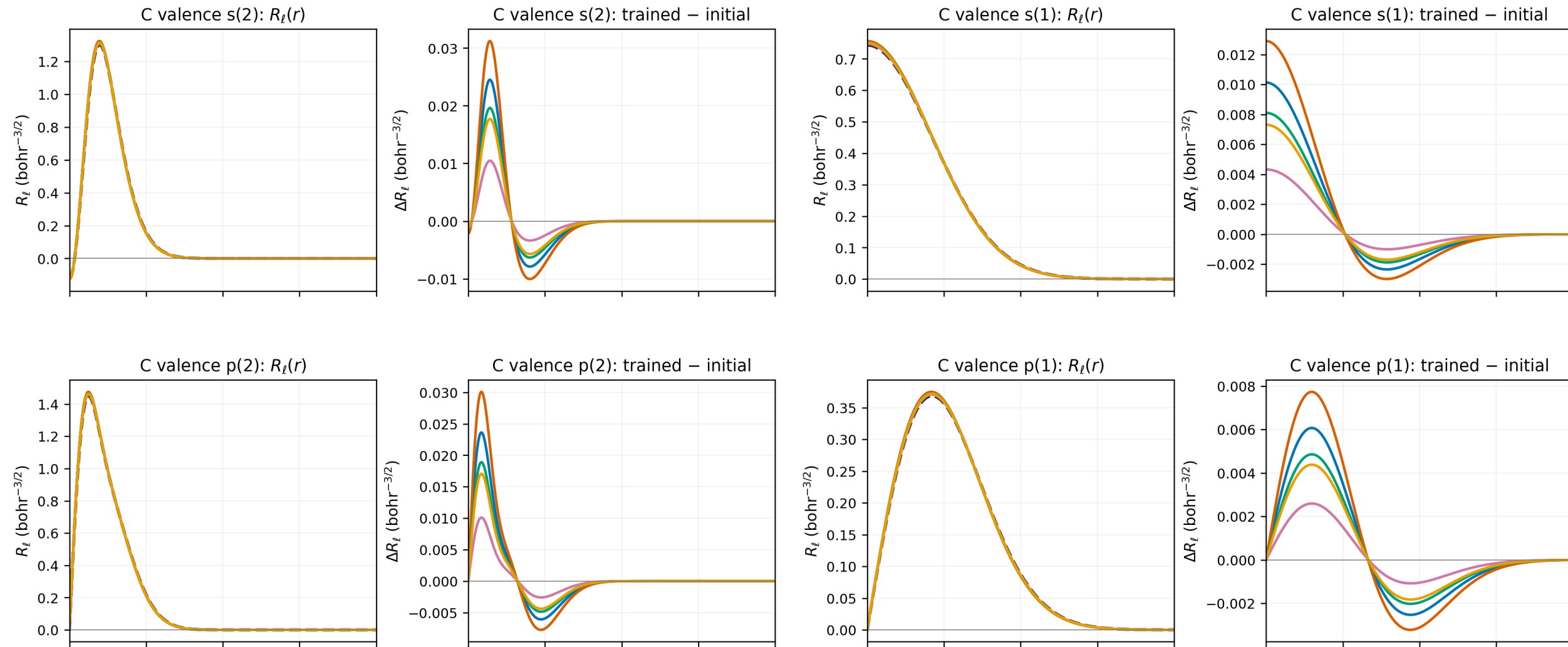


Chemical Environment-Adaptive Basis

➤ 1: Neural Network Atom Basis Designing and Training

— initial (all PES points) — grid (0, 14): (1.330, 1.120) Å — grid (14, 0): (1.450, 1.040) Å
— grid (0, 0): (1.330, 1.040) Å — PES minimum: (1.381, 1.074) Å — grid (14, 14): (1.450, 1.120) Å

Dashed line: zero-head initial basis. Colored lines: epoch-100 basis. C core s(3) is fixed and omitted.

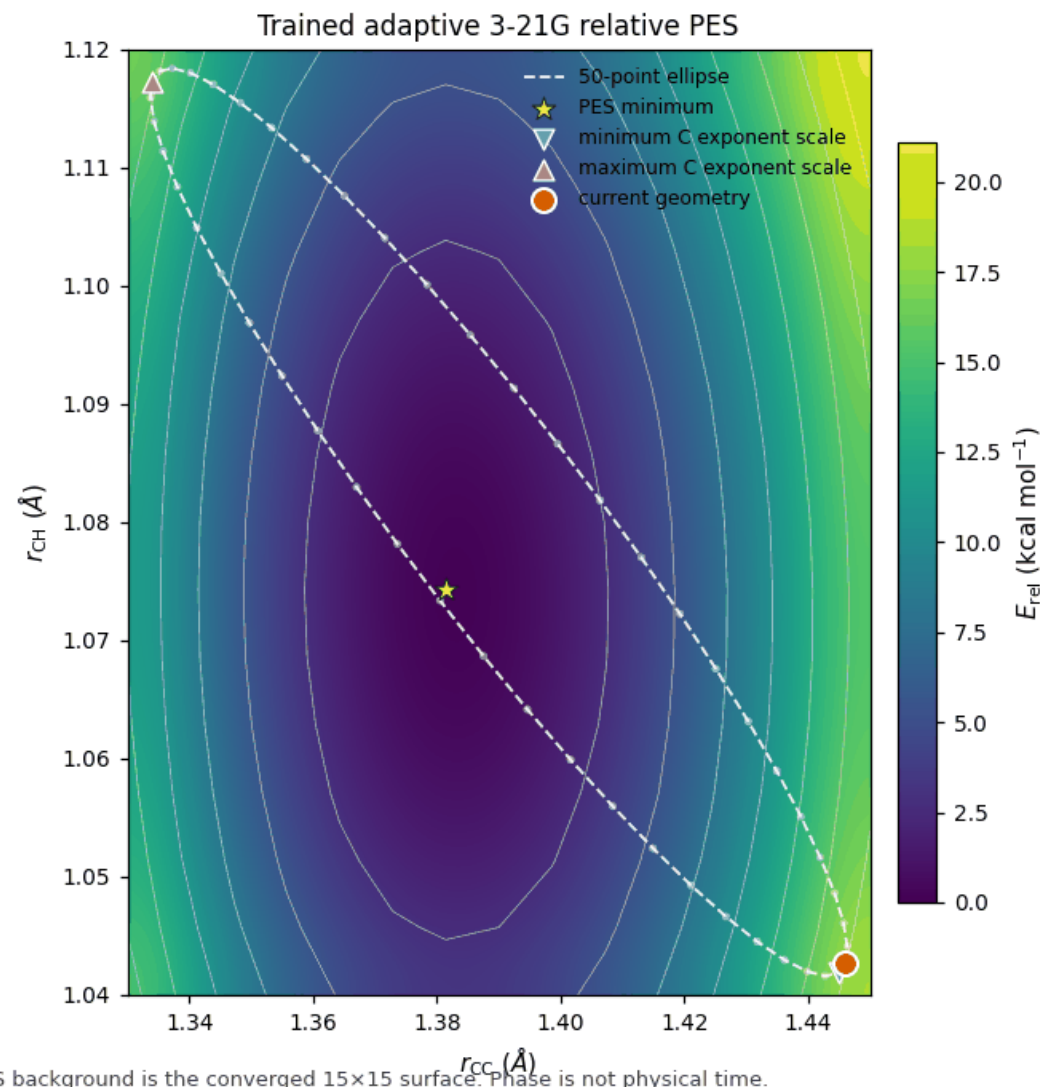
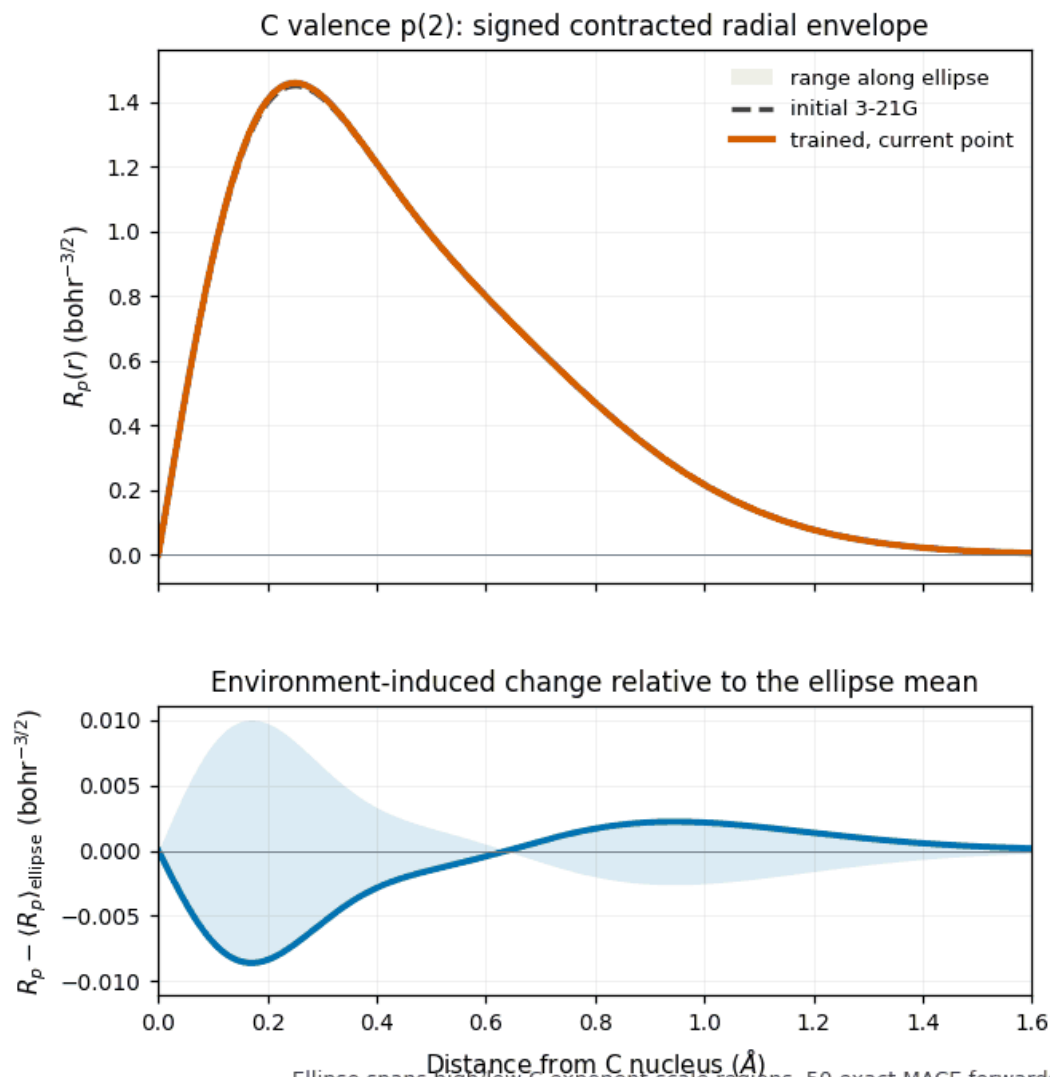


■ Chemical Environment-Adaptive Basis

➤ 1: Neural Network Atom Basis Designing and Training

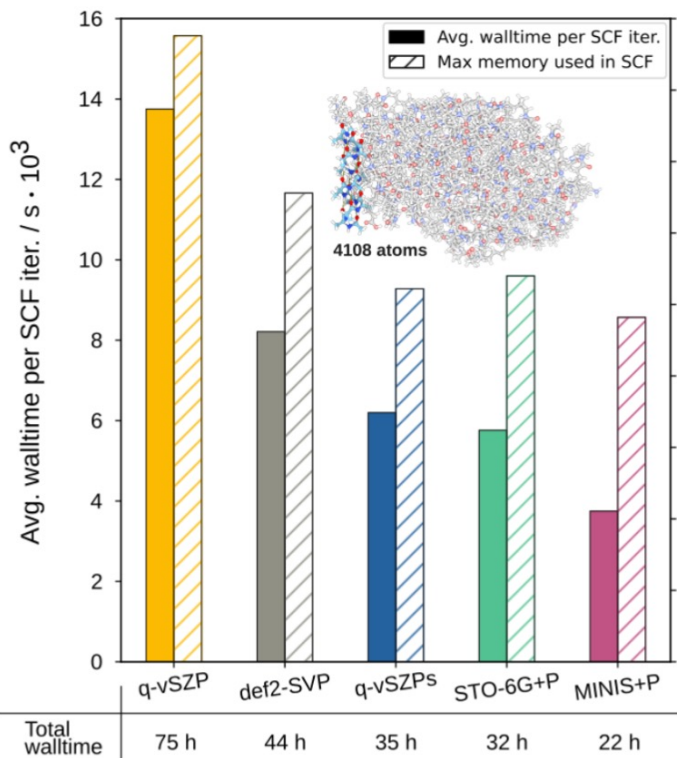
Environment-adaptive C valence radial function along a contrast ellipse

frame 01/50 | $r_{CC}=1.44600 \text{ \AA}$ | $r_{CH}=1.04267 \text{ \AA}$ | $\lambda_C=1.004045$ | $\alpha'/\alpha=1.008107$



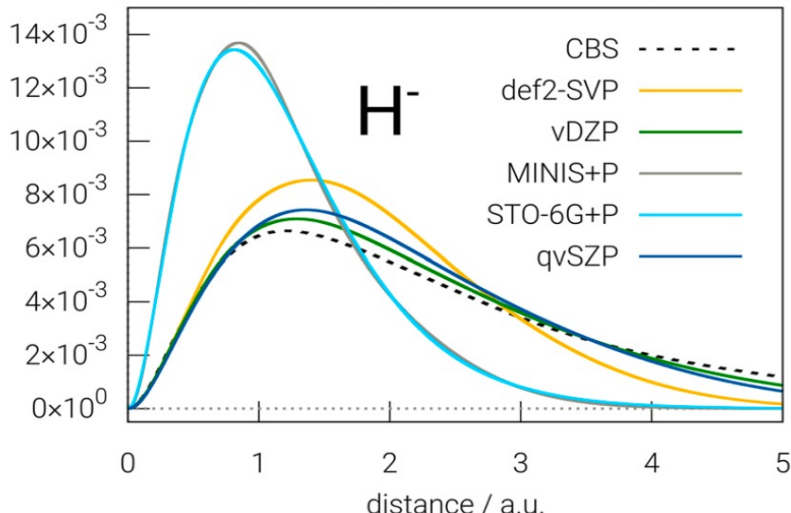
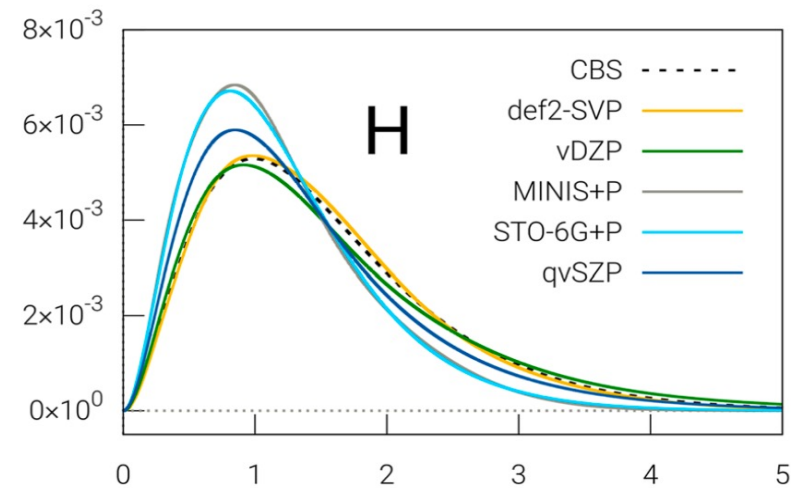
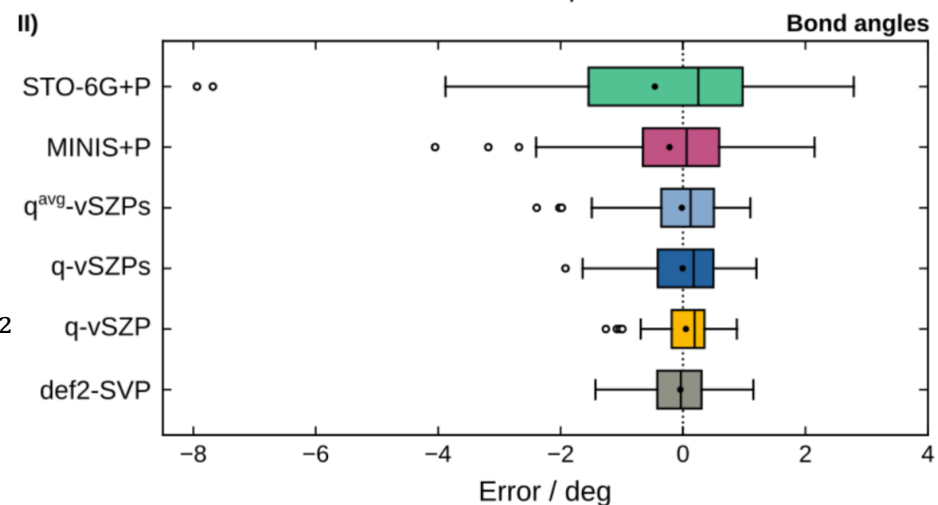
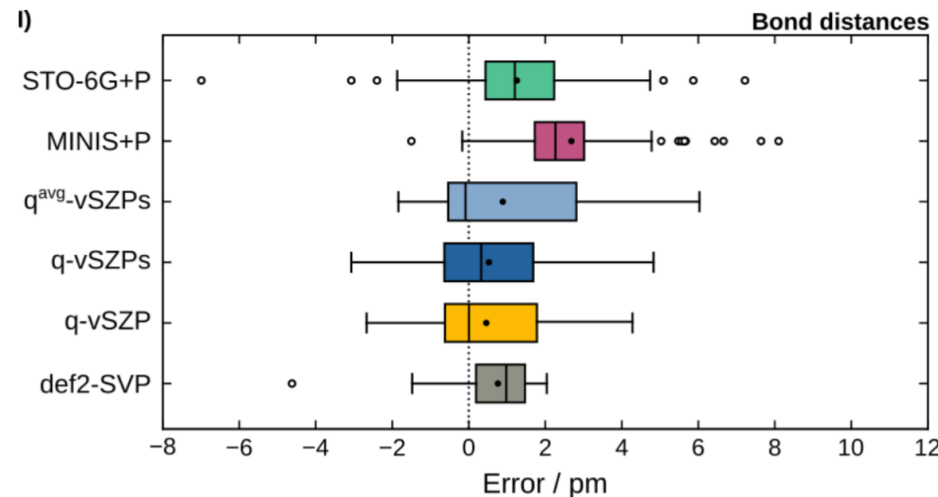
Ellipse spans high/low C exponent-scale regions. 50 exact MACE forwards; PES background is the converged 15×15 surface. Phase is not physical time.

Chemical Environment-Adaptive Basis q-vSZP/q-vSZPs



$$R_{A\ell}(\mathbf{r}; q_A^{\text{eff}}) = \sum_p C_{p,A}(q_A^{\text{eff}}) N_{\ell p} r^\ell e^{-\alpha_{\ell p} r^2}$$

$$C_{p,A}(q_A^{\text{eff}}) = C_{p,A}(q_A^{\text{eff}}) + C_{p,0}$$



J. Chem. Phys. 165, 054109 (2026)

J. Chem. Phys. 159, 164108 (2023)

■ Chemical Environment-Adaptive Basis EGNN for Contraction Coefficients

➤ SO(3) Convolution Filter

$$F_m^{l_{in}l_f}(\mathbf{r}_{ij}) = R^{l_{in}l_f}(|\mathbf{r}_{ij}|)Y_m^{l_f}\left(\frac{\mathbf{r}_{ij}}{|\mathbf{r}_{ij}|}\right)$$

$$Y_m^{l_f}\left(\frac{\mathbf{r}_{ij}}{|\mathbf{r}_{ij}|}\right) \rightarrow \text{Harmonic Spherical Function}$$

$$R^{l_{in}l_f}(|\mathbf{r}_{ij}|) \rightarrow \text{Feed Forward Network}$$

➤ SelfNet Layer

$$f_{ii}^{l_{out}} = \left(W^{l_{in}r,l_{in}l,l_{out}} x_i^{l_{in}} \otimes x r_i^{l_{in}}\right)^{l_{out}}$$

$$m_{ijc}^{l_{out}} = \sum_{ll} \left(F_c^{l_{in}l_f,l_{out}}\left(r_{ij}, \frac{r_{ij}}{|\mathbf{r}_{ij}|}\right) \otimes \left(x_{jc}^{l_{in}}\right)\right)^{l_{out}}$$

➤ PairNet Layer

$$f_{ij}^{l_{out}} = \left(F^{l_{in}r,l_{in}l,l_{out}}(\mathbf{r}_{ij}) x_i^{l_{in}i} \otimes x_j^{l_{in}j}\right)^{l_{out}}$$

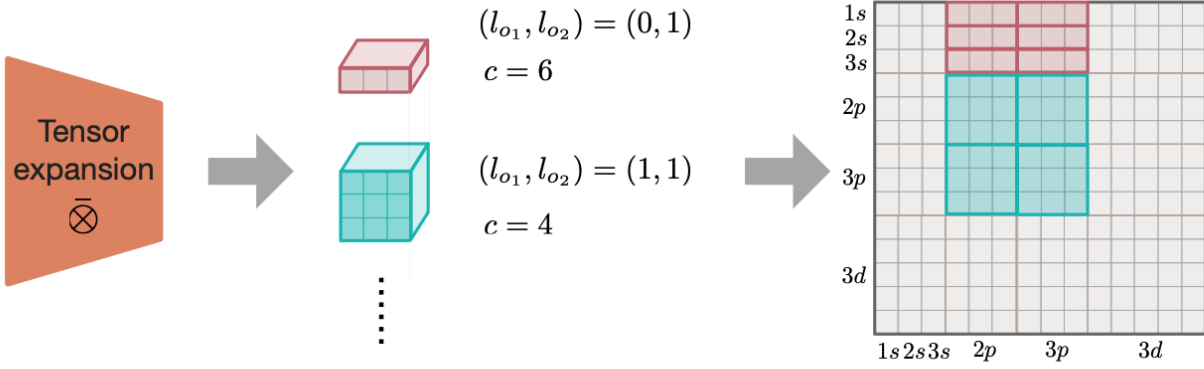
$$\tilde{x}_i = f_{linear}(x_i + \sum_j m_{ij})$$

➤ Expansion Layer

$$\left(\overline{\otimes} f_{ij}^{l_3}\right)_{m_1,m_2}^{l_1,l_2} = \sum_m C_{l_1 m_1, l_2 m_2}^{l_3 m_3} f_{ij}^{l_3}$$

Hamiltonian Learning 

Neural Network Basis Set 



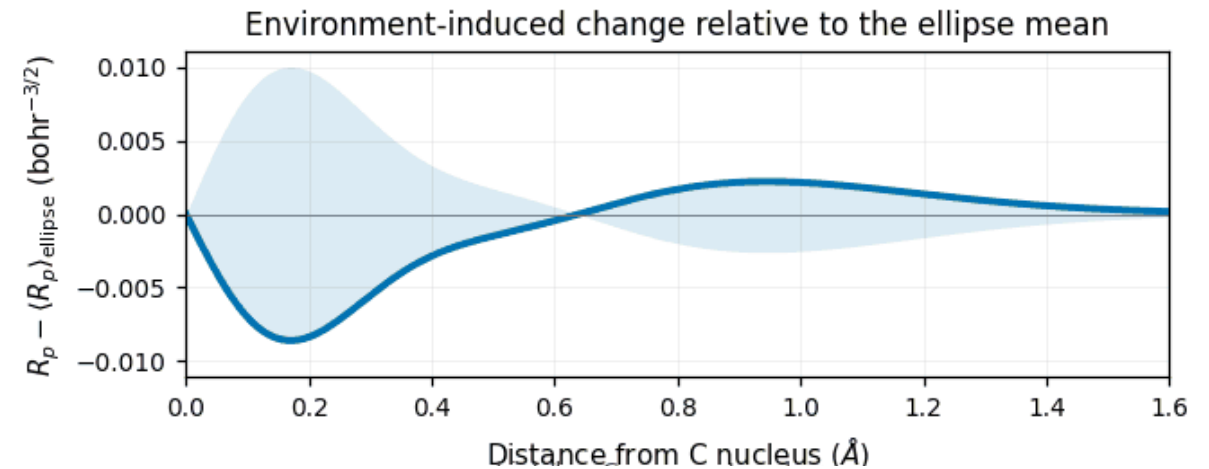
■ Functional

● Functional-GradTDDFT

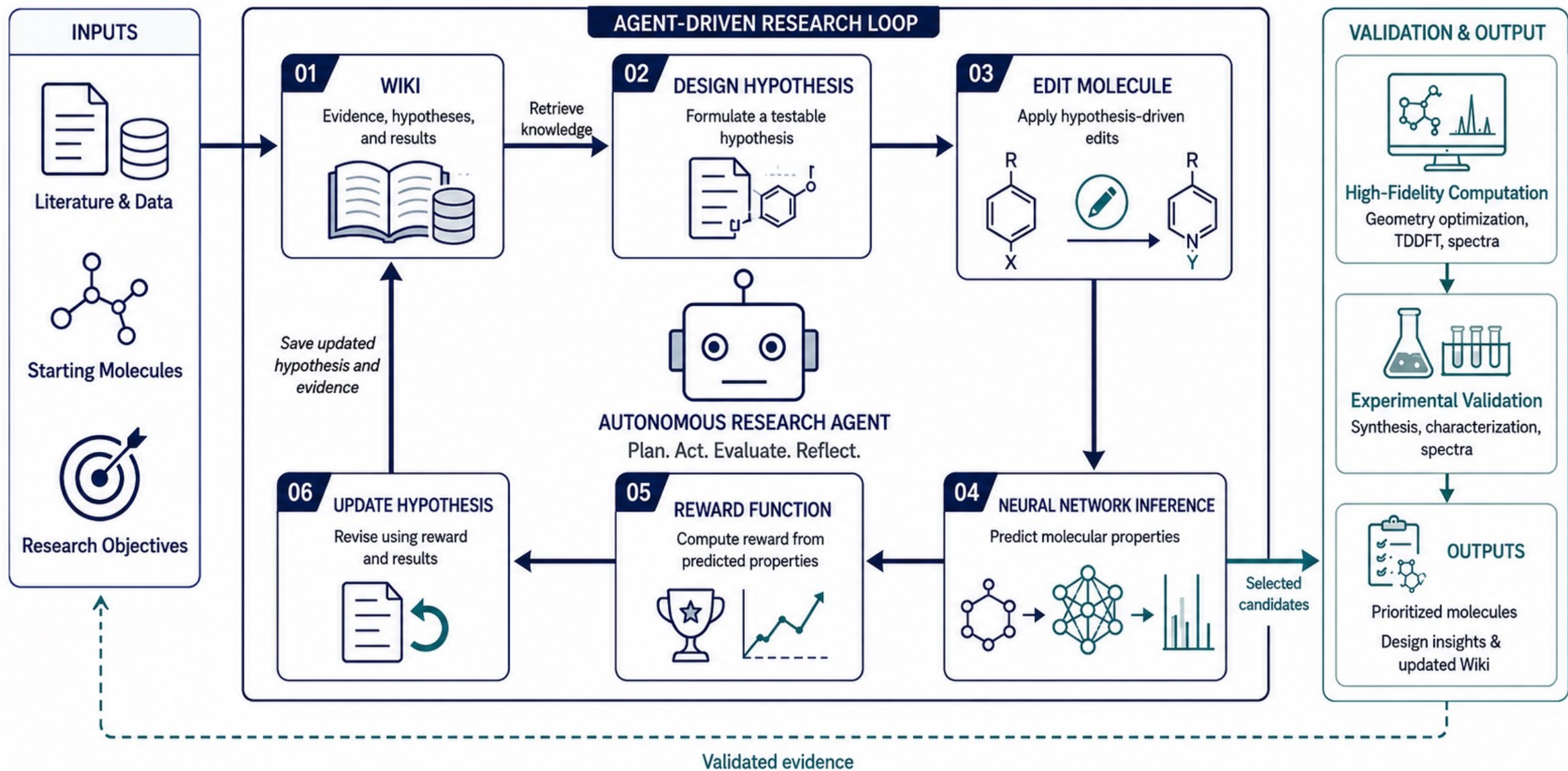
- Differential SCF code for DFT/HF and implicit differential Davidson solver code for TDDFT
 - Basis Set Library and Electronic Integral Module With Jax Based Contraction Operator
 - Neural Network for Optimization of Gaussian-Type Orbital
-

● Outlook

- Highly Efficient Integral Module for AD
- SE(3) Equivariant Neural Network Basis Set
- Neural Network Based NAO(More Flexible Form)
- Form GradTDDFT to GradSCF (Electronic Structure Infra for AI)



Background - Molecular Designing



■ Molecular Edition Protocol

➤ Basic Operation

■ Atom operations

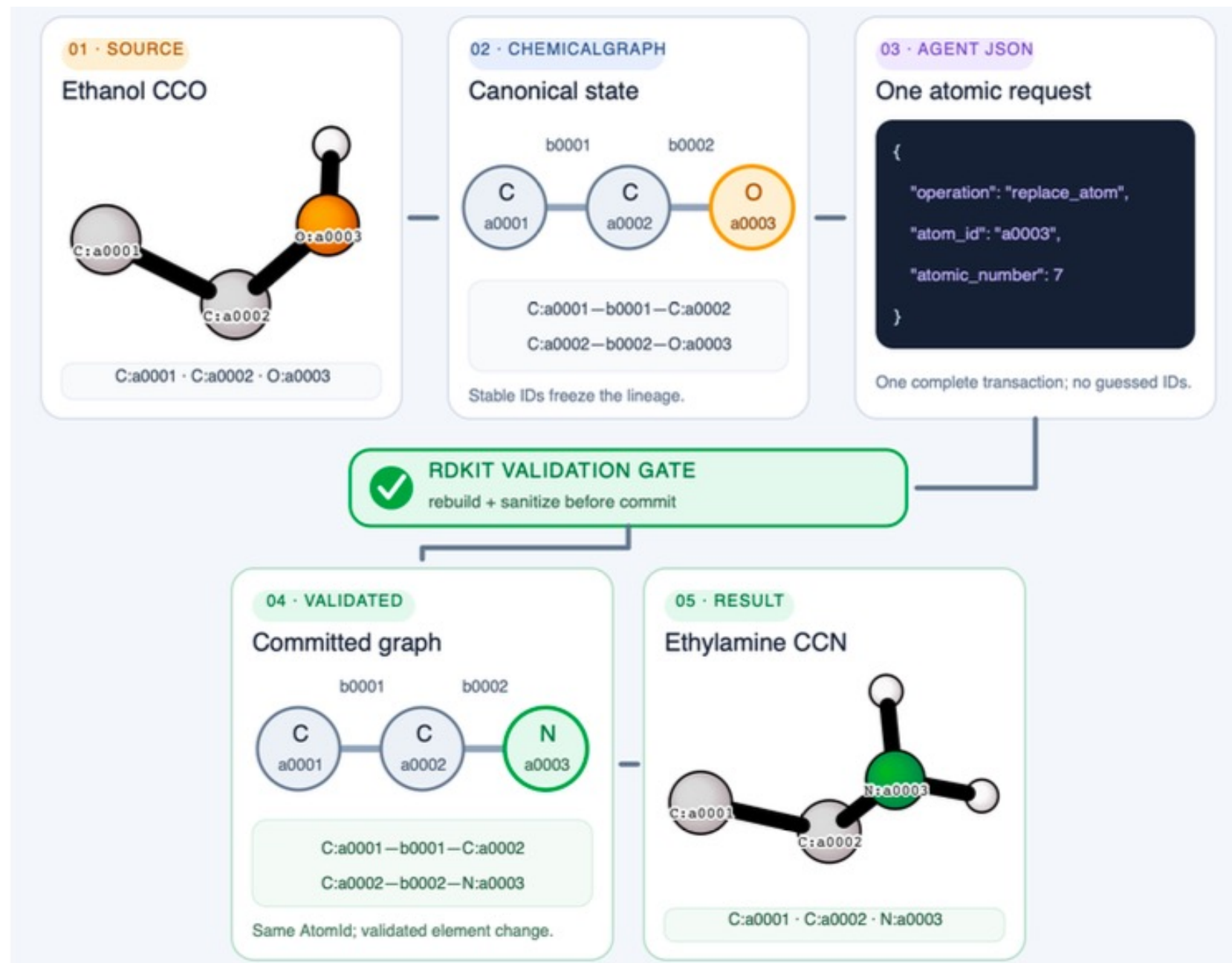
- Add Atom
- Remove Atom
- Replace Atom

■ Bond operations

- Add Bond
- Remove Bond
- Change Bond

■ Fragment operations

- Attach Fragment
- Detach Fragment
- Substitute Fragment



Molecular Edition Protocol

Benzene -> all recorded attempts | frame 1/101 | CALCULATED

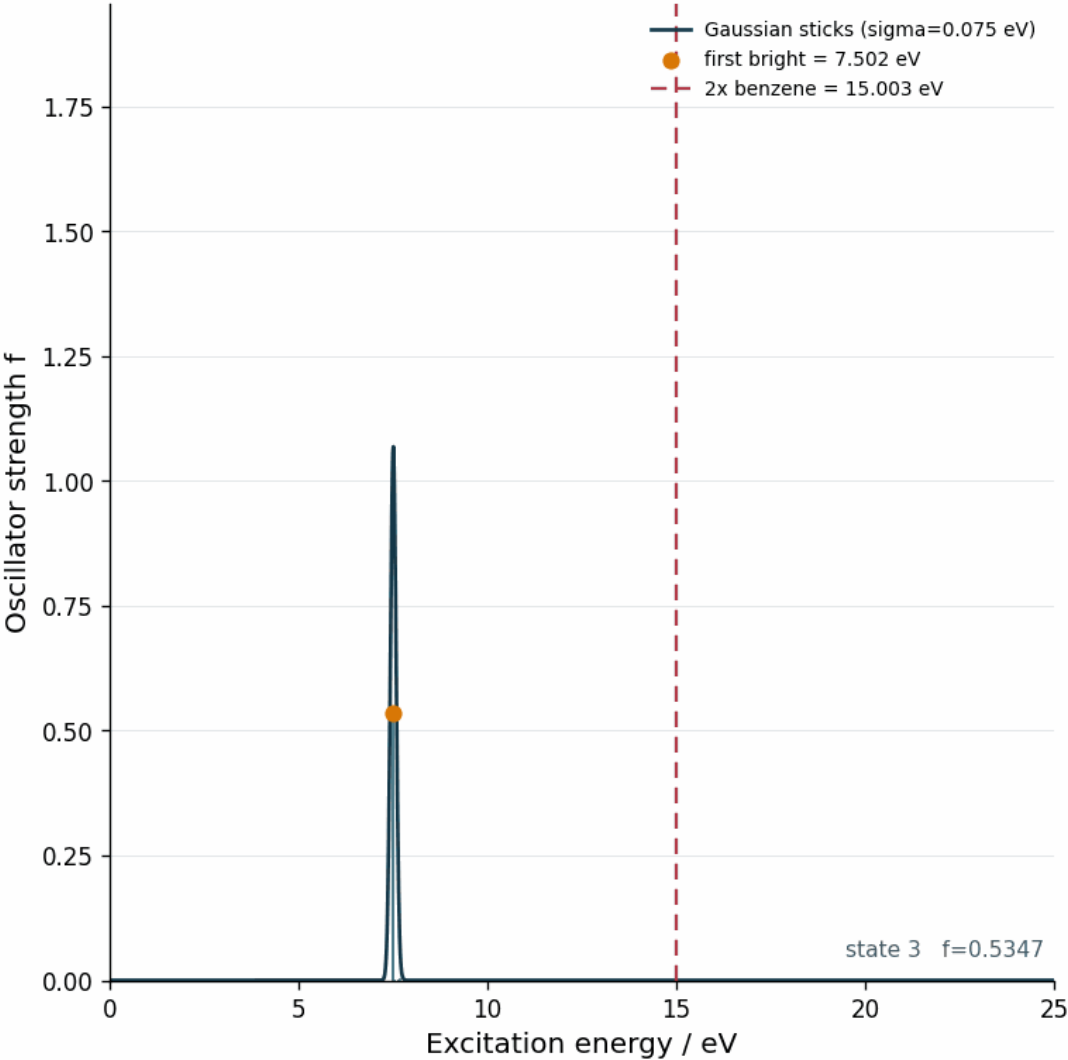
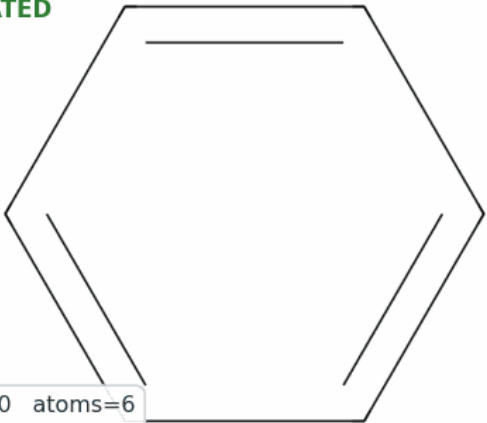
PBE/3-21G TDDFT spectrum

Molecule | MAIN LINE

parent: step-000-benzene

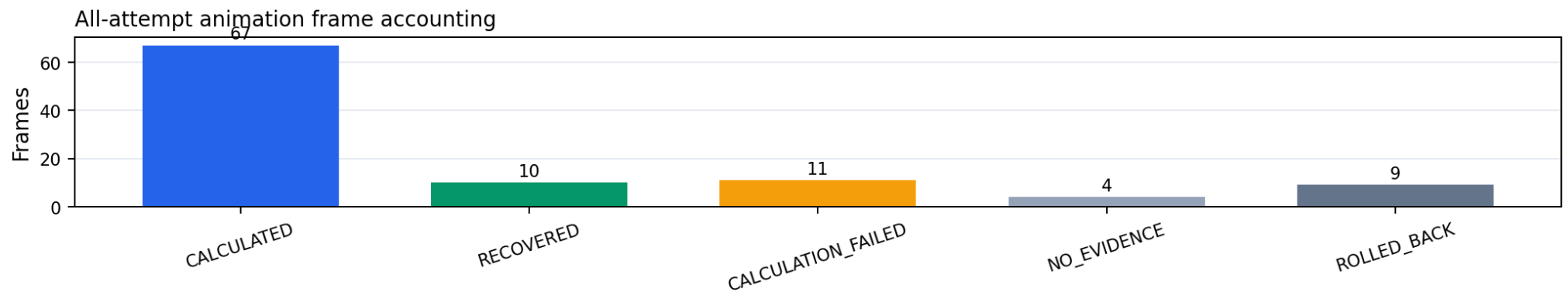
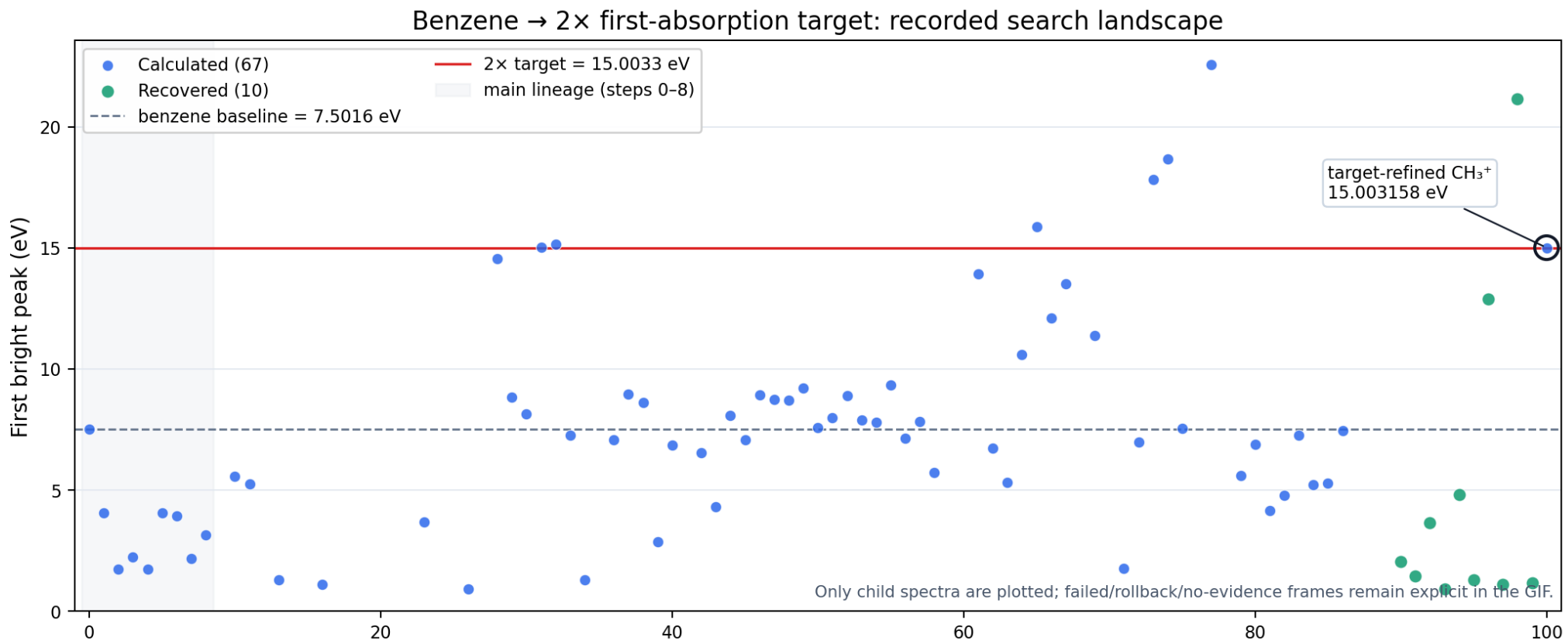
CALCULATED

C6H6 q=0 atoms=6
benzene
edit: initial_structure



Every frame is a recorded attempt; branch parent is explicit. Failed or rolled-back children do not receive invented spectra.

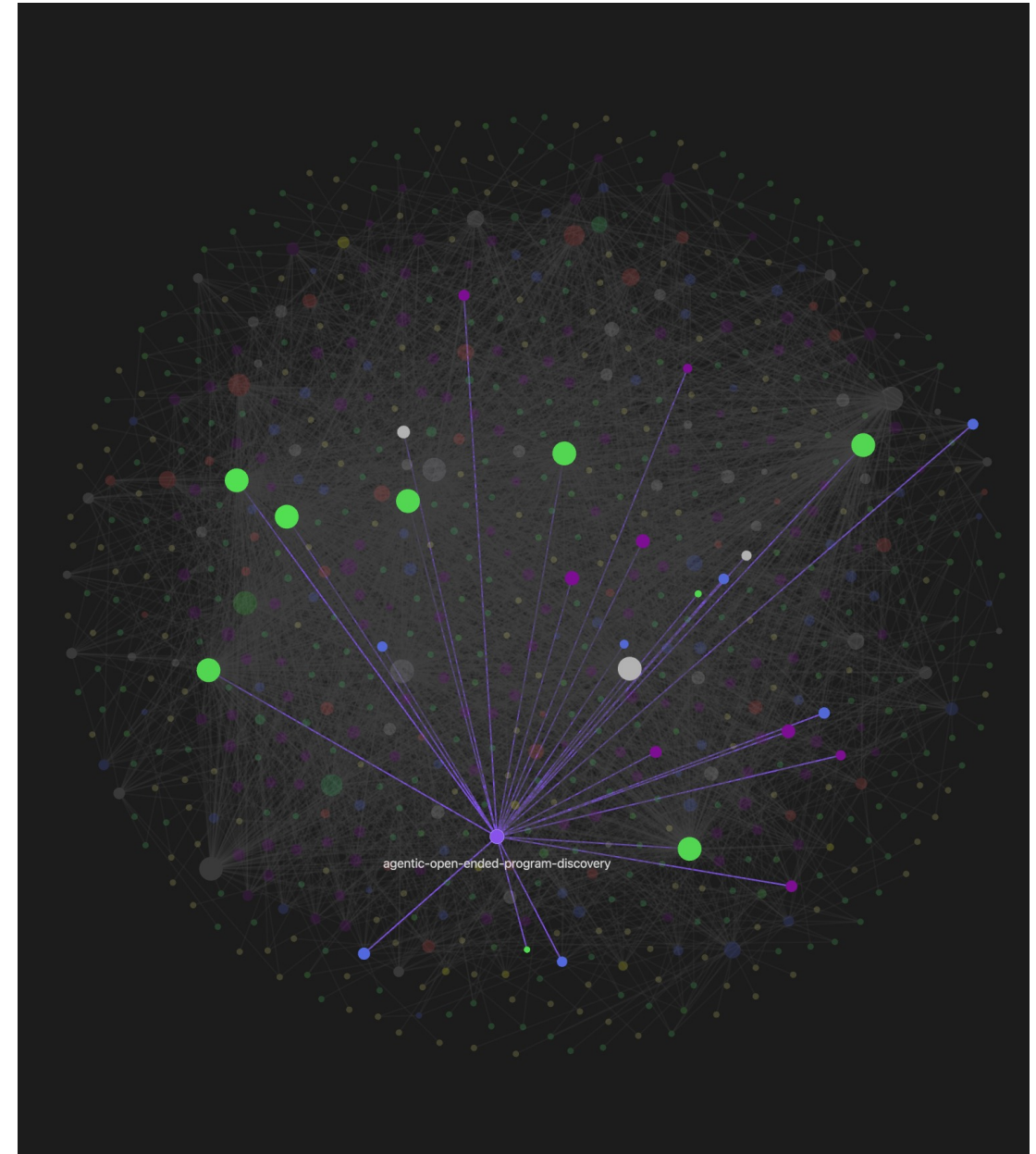
Molecular Edition Protocol



Hypothesis from LLM-Wiki (BTW)

```
.wiki/  
├── raw/          immutable local sources; never published by this repository  
├── wiki/  
│   ├── sources/  
│   ├── concepts/  
│   ├── topics/  
│   ├── queries/  
│   ├── relations.md  
│   └── method-improvements.md  
└── output/      derived local artifacts
```

```
PDF reader + existing wiki  
    │  
    v  
Evidence Pack  
    │  
    +--> Organizer: source draft  
    │  
    +--> Questioner: 3 questions by default  
           │  
           v  
Round 1: Organizer answers  
        -> Evaluator scores/feedback  
Round 2: Organizer revises  
        -> Evaluator scores/feedback  
Round 3: Organizer revises  
        -> Evaluator final score  
           │  
           v  
Main Agent validates and writes:  
source page + agent-queries page + relations/topic/index/log
```



■ Agent for Hypothesis, Molecular Edition and Evolution

Parent molecule

CC(=O)C=Cc1c(N(C)C)cc2c(=O)c3c(C=CC=O)cccc3n3c4ccccc4c(=O)c1c23

The parent contains two conjugated peripheral arms:

- an acetyl-vinyl arm, CC(=O)C=C-;
- an aldehyde-vinyl arm, -C=CC=O.

The proposed edit retains the first arm and replaces the second one.

Hypothesis

Replace the aldehyde-vinyl arm -C=CC=O with a stronger conjugated dicyanovinyl acceptor, -C=C(C#N)C#N, while preserving the fused polycyclic chromophore and the existing dimethylamino donor.

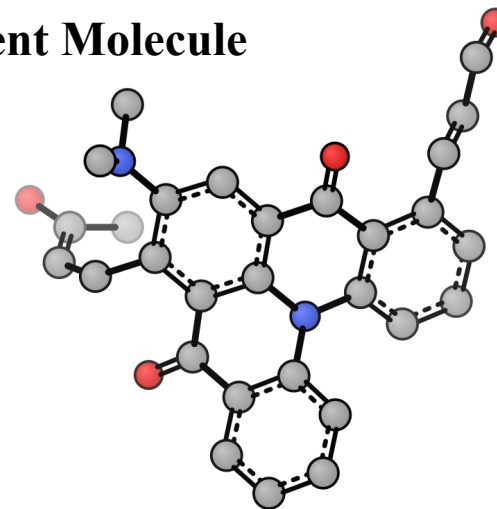
The two cyano groups are expected to strengthen the electron-withdrawing character of the side chain. Coupling this acceptor to the π -conjugated core may increase donor-acceptor polarization, stabilize the relevant excited state, reduce the effective optical gap, and red-shift the first absorption maximum toward the target range of 620–750 nm.

The hypothesis is falsified at two levels:

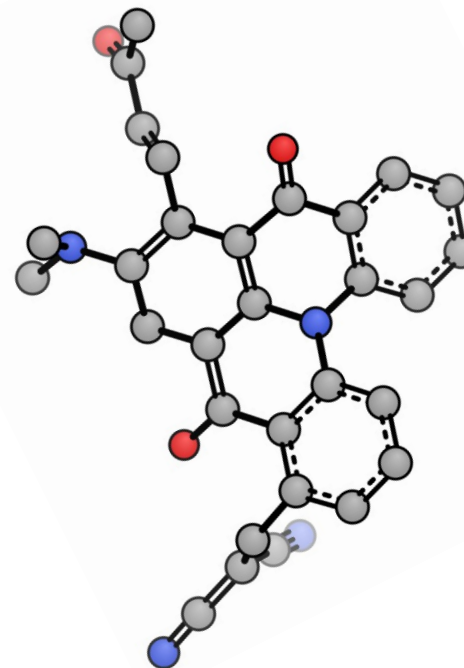
1. Directional falsification: the absorption maximum does not shift to a longer wavelength.
2. Target-level falsification: the absorption maximum red-shifts but remains substantially below 620 nm.

PLQY and $\log_{10}(\epsilon)$ remain weak secondary objectives and are not required to improve simultaneously.

Parent Molecule



Edited Molecule



■ Agent for Hypothesis, Molecular Edition and Evolution

Proposed molecular edit

Parent side chain:
core-C=CH-CHO

substitute_fragment

↓

Designed side chain:
core-C=C(C#N)₂

More explicitly:

...c3c(C=CC=O)cccc3...

↓

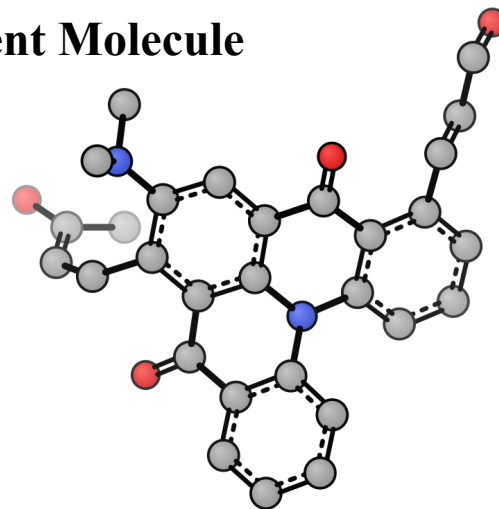
...c3c(C=C(C#N)C#N)cccc3...

This is a **substitute_fragment** operation rather than a single-atom replacement:

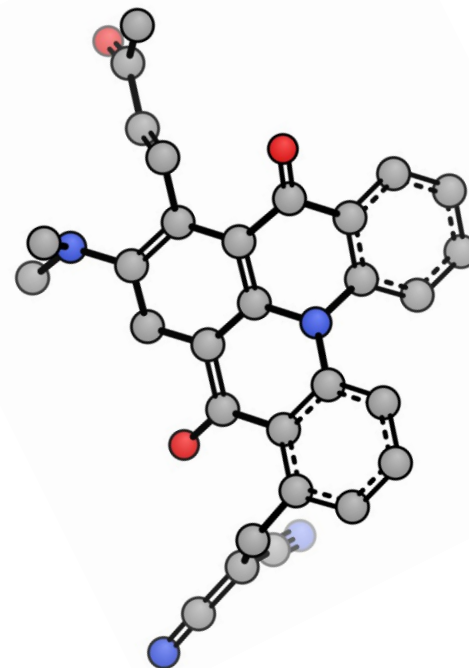
Operation:	substitute_fragment
Cleaved bond:	b0028
Retained parent atom:	a0024
New attachment bond:	DOUBLE
Added fragment:	C3N2
Complete edited arm:	C=C(C#N)C#N

The complete dicyanovinyl arm contains six heavy atoms, one of which is retained from the parent structure; the newly attached fragment contributes five heavy atoms.

Parent Molecule



Edited Molecule

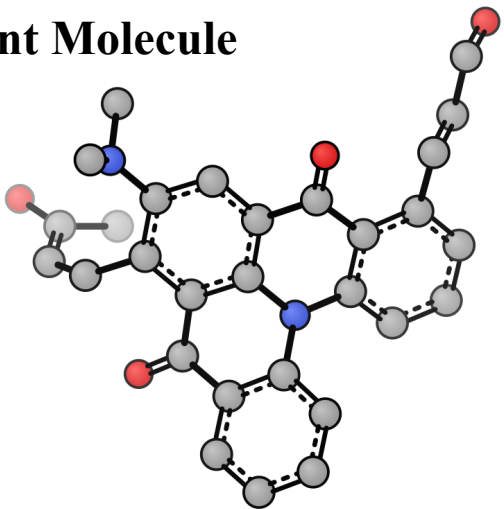


■ Agent for Hypothesis, Molecular Edition and Evolution

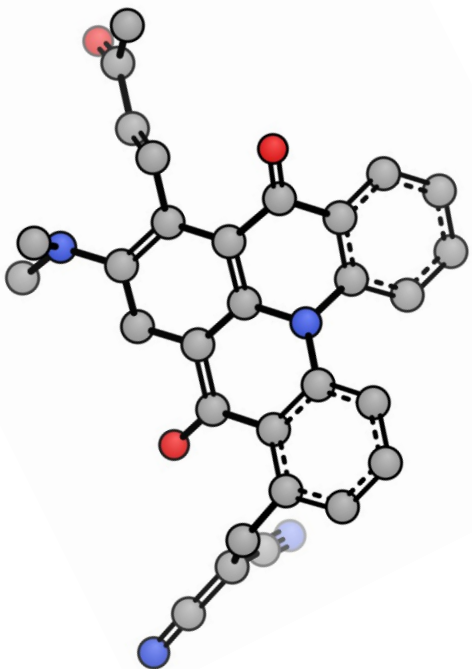
Results

Property	Parent	Edited molecule	Change
Absorption maximum, λ_{abs}	501.981 nm	525.261 nm	+23.280 nm
Distance from 620 nm	118.019 nm	94.739 nm	Improved by 23.280 nm
Emission maximum, λ_{emi}	483.260 nm	478.009 nm	-5.251 nm
PLQY	0.2088	0.0976	Decreased
Molar extinction coefficient	26,366.4 M ⁻¹ cm ⁻¹	29,303.6 M ⁻¹ cm ⁻¹	Increased
log10(ϵ)	approximately 4.421	4.467	Slight increase
Fitness	-0.904427	-0.725828	Improved
Parent similarity	-	0.7385	Main scaffold retained

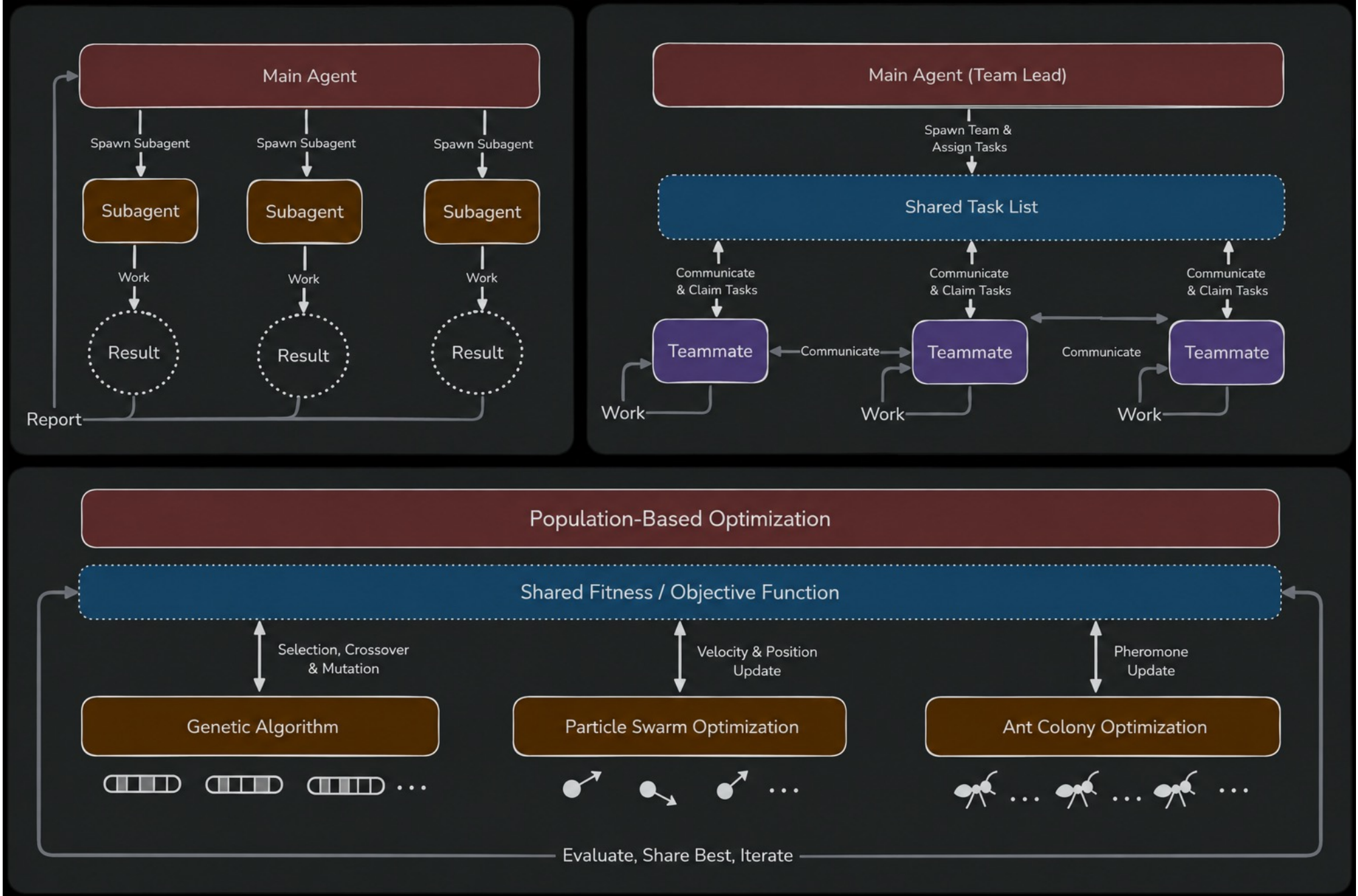
Parent Molecule



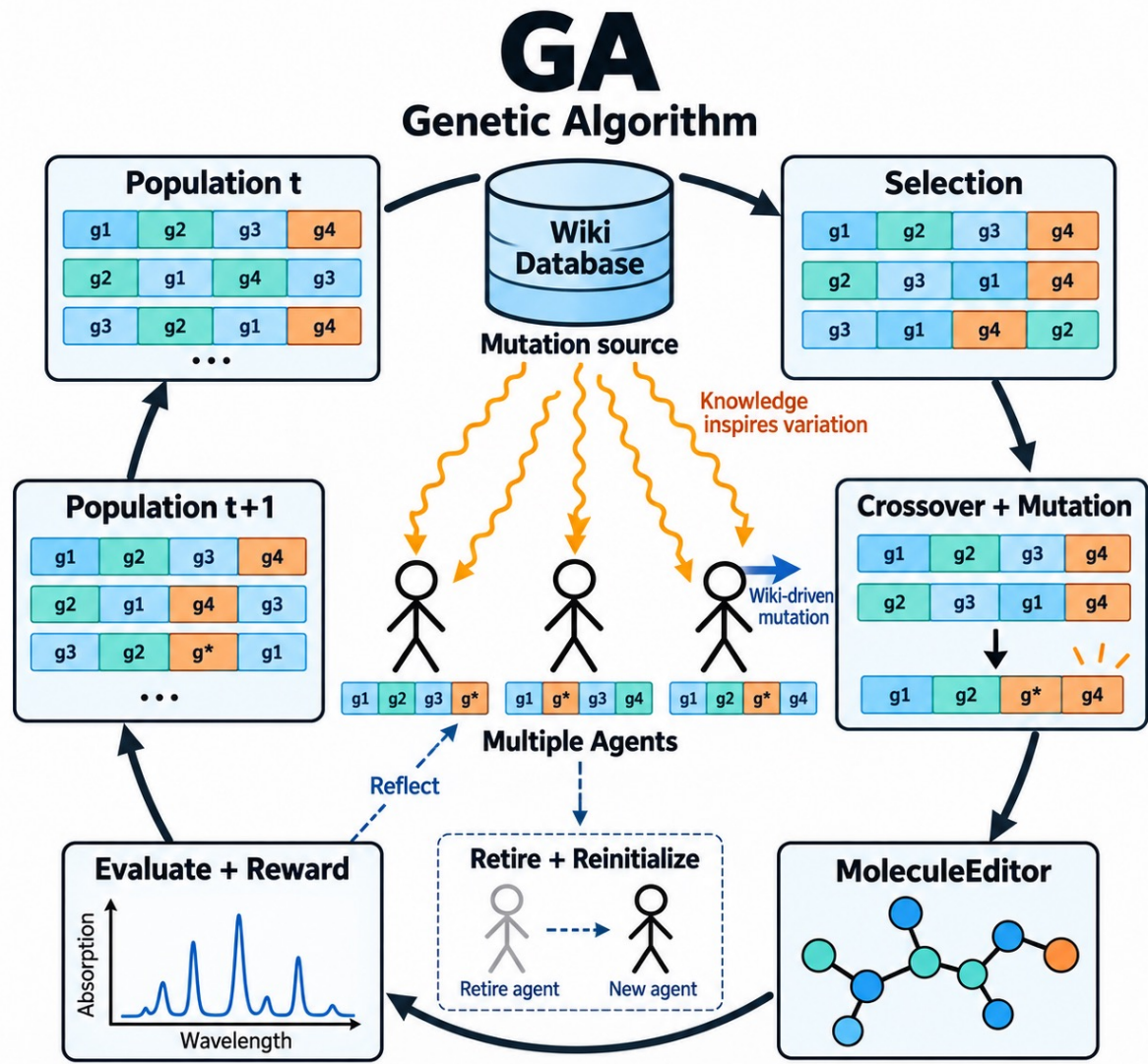
Edited Molecule



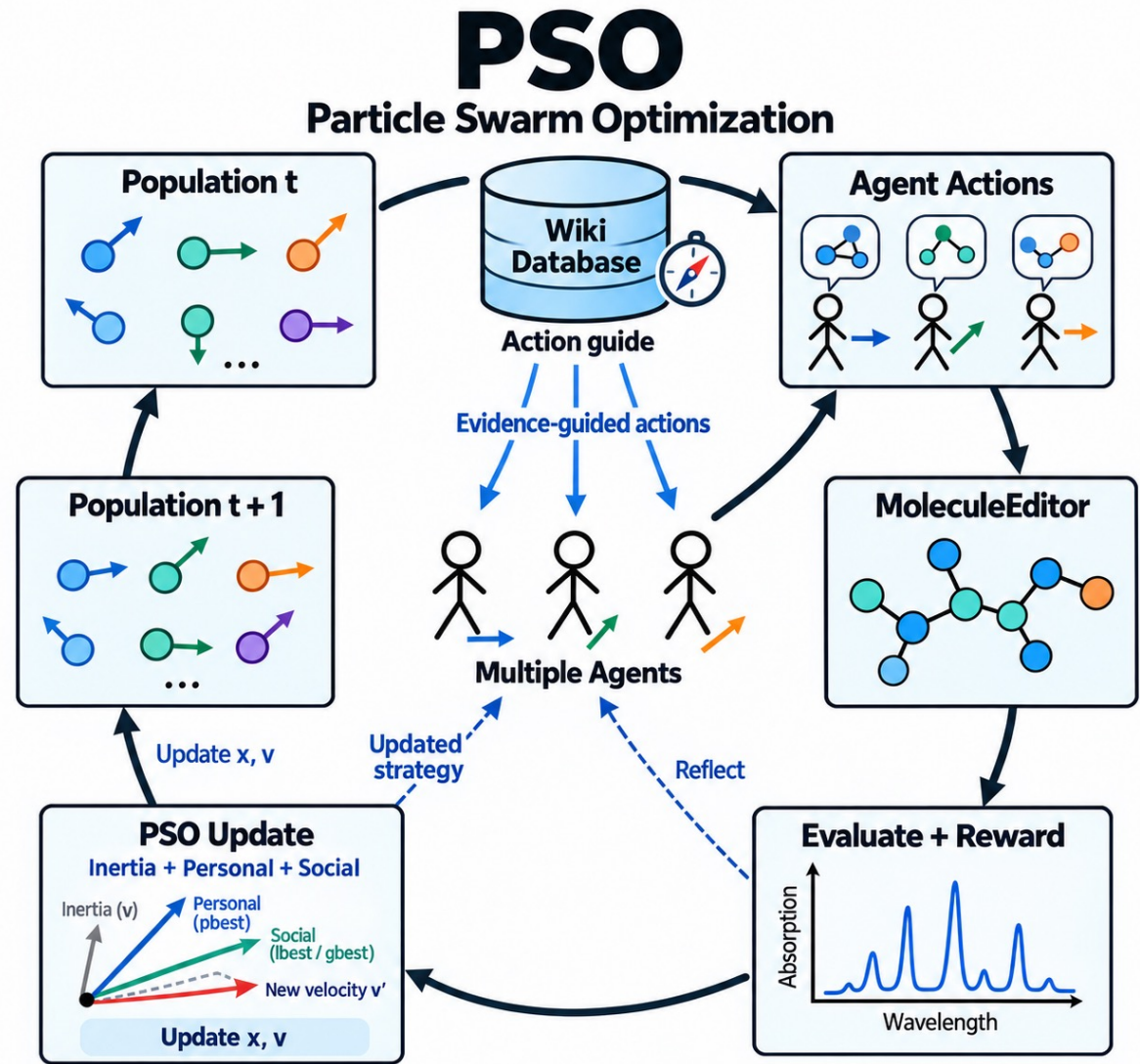
■ Agent for Hypothesis, Molecular Edition and Evolution



■ Agent for Hypothesis, Molecular Edition and Evolution

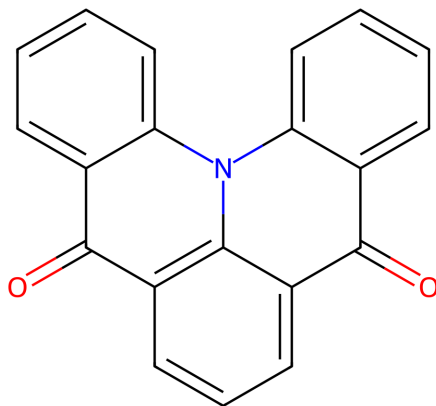


Genes = edit operations · Phenotype = molecule



Position = edit strategy · Velocity = strategy change

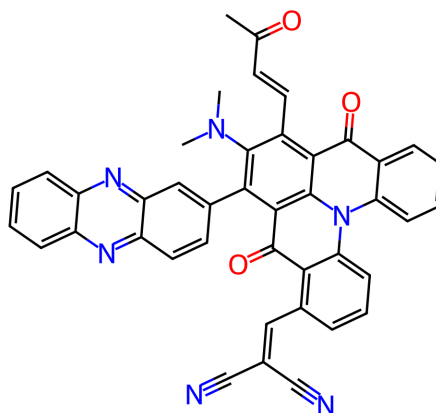
■ Agent for Hypothesis, Molecular Edition and Evolution-PSO



Initial DiKTA

Absorption: 432.94 nm

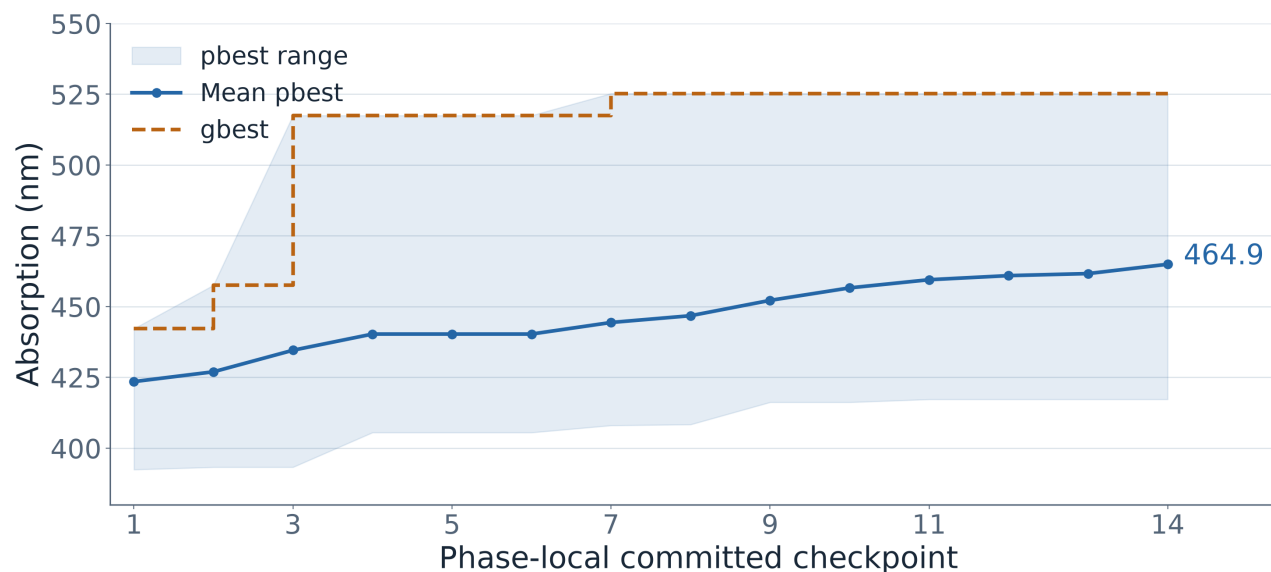
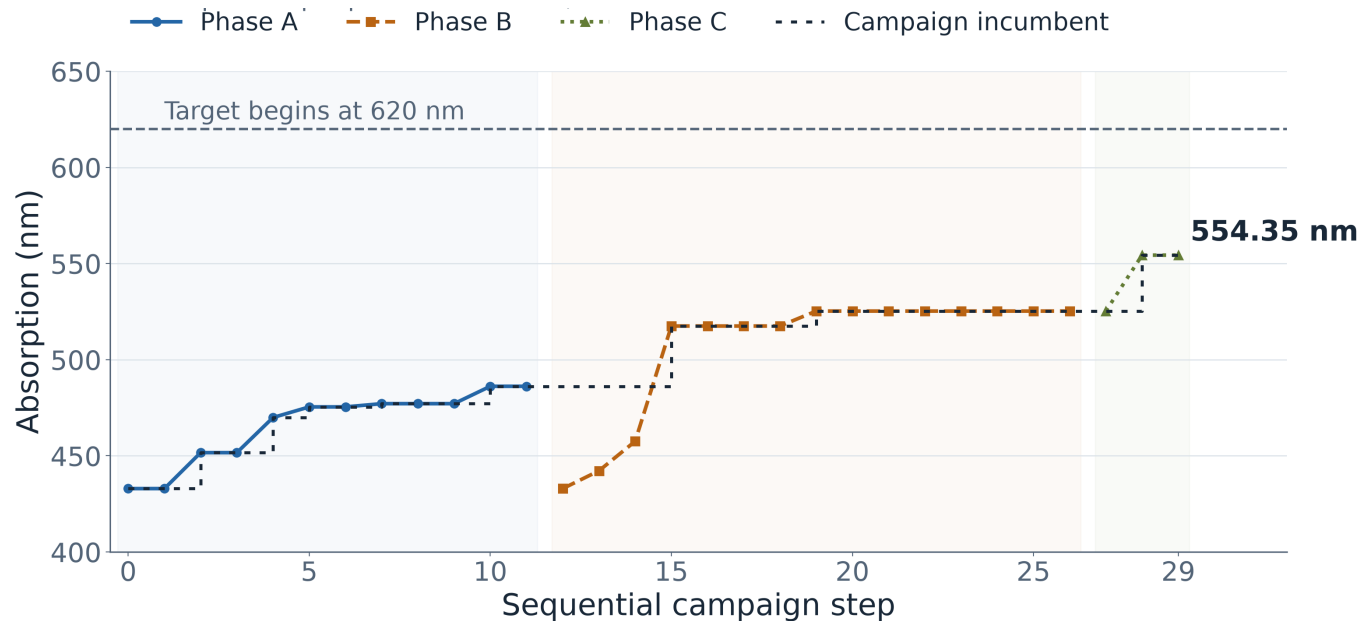
Reward: -1.434828



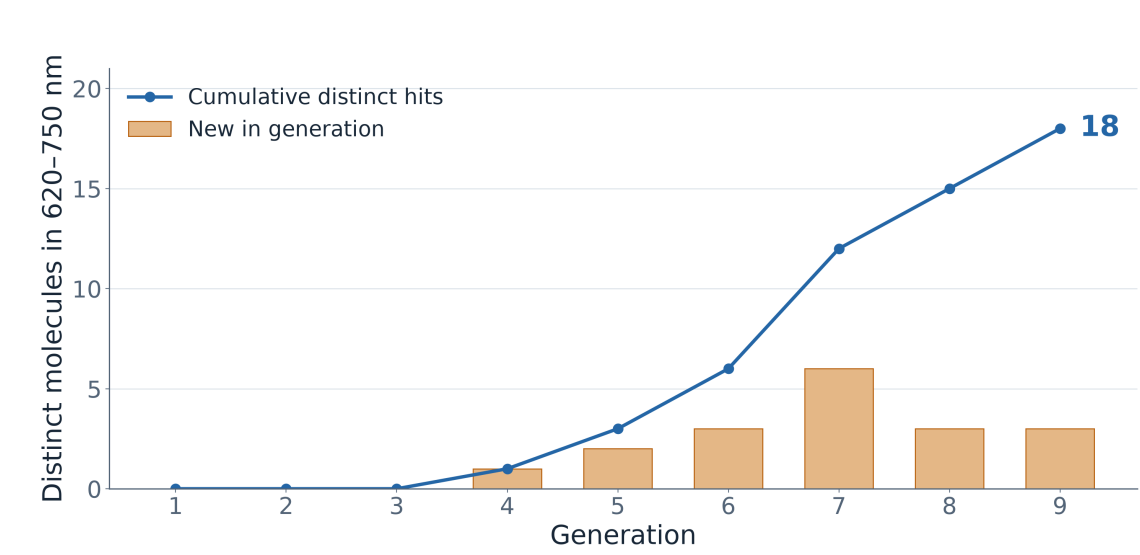
Phase C endpoint

Absorption: 554.35 nm

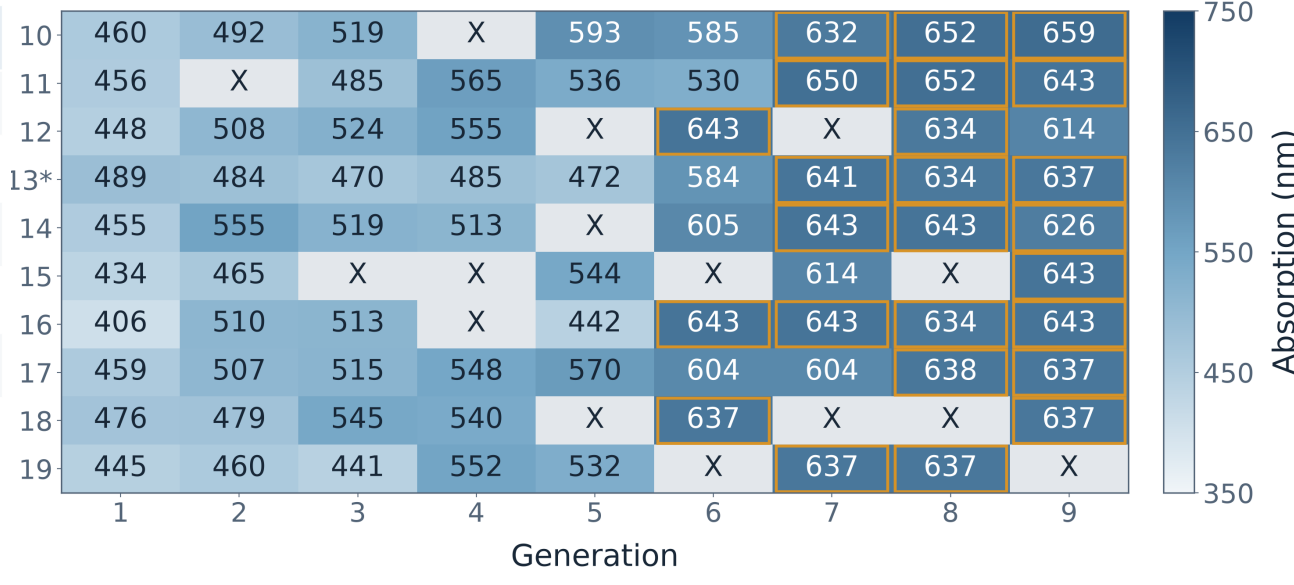
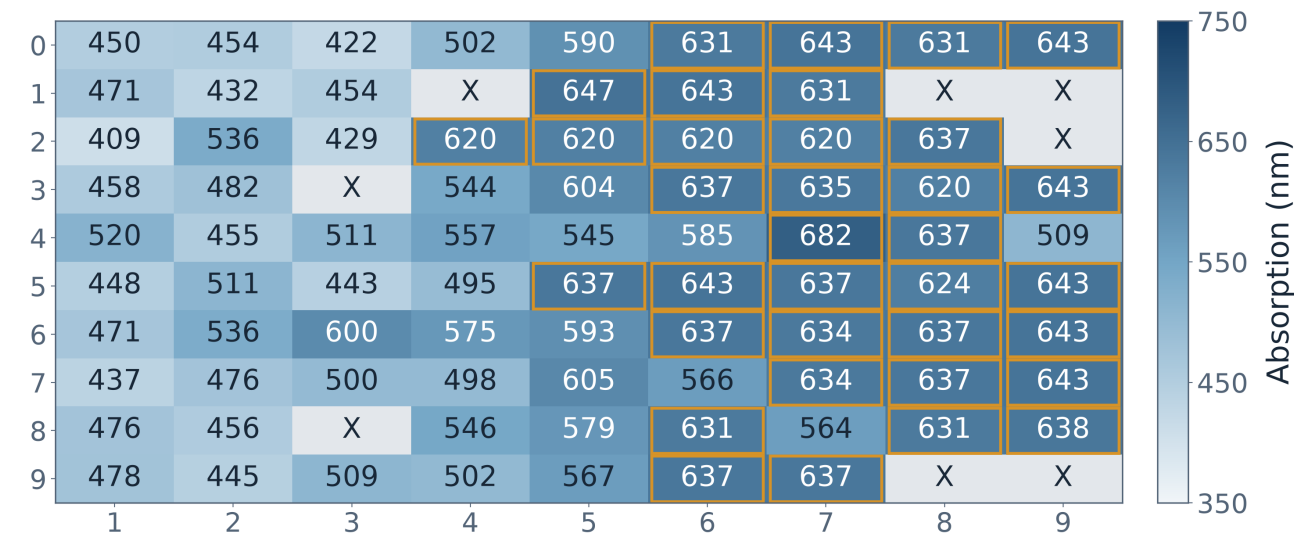
Reward: -0.502542



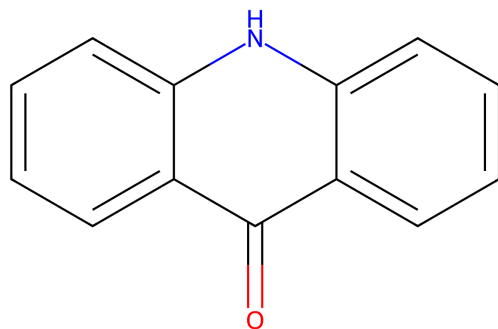
Agent for Hypothesis, Molecular Edition and Evolution-GA



Rank	Abs. (nm)	Emi. (nm)	PLQY	log ₁₀ ε	Reward
1	620.32	669.59	0.535	4.296	1.004832
2	635.15	667.76	0.590	4.123	1.004824
3	643.46	658.90	0.542	4.001	1.004378
4	638.47	670.15	0.546	3.950	1.004312
5	650.32	671.72	0.514	4.018	1.004268
6	636.71	645.07	0.431	4.239	1.004222



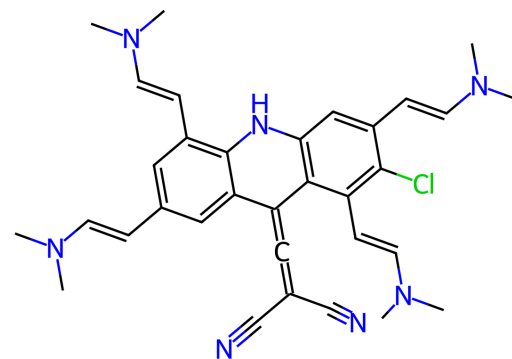
■ Agent for Hypothesis, Molecular Edition and Evolution-GA



Initial acridone

Absorption: 393.30 nm

Reward: -1.741050

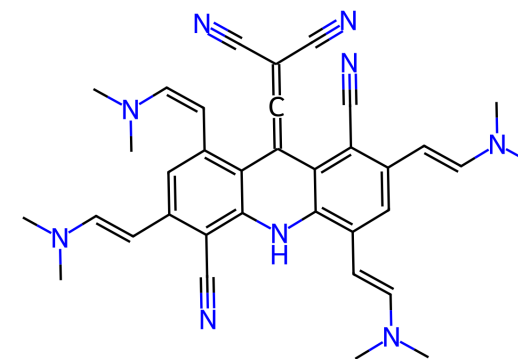


GA-09

Absorption: 638.30 nm

Emission: 678.71 nm | PLQY: 0.535

ϵ : 8102 M⁻¹ cm⁻¹ | Reward: 1.004189

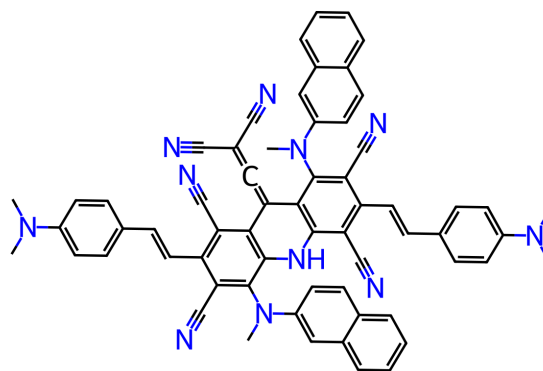


GA-10

Absorption: 651.91 nm

Emission: 693.82 nm | PLQY: 0.486

ϵ : 11239 M⁻¹ cm⁻¹ | Reward: 1.004179

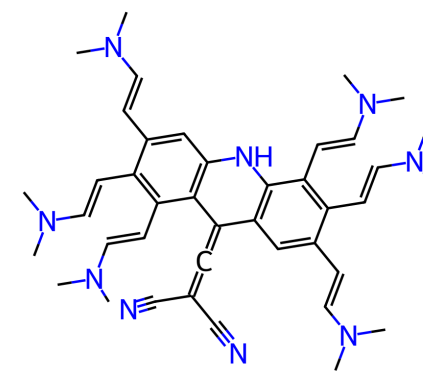


GA-17

Absorption: 646.62 nm

Emission: 772.08 nm | PLQY: 0.194

ϵ : 40431 M⁻¹ cm⁻¹ | Reward: 1.003646



GA-18

Absorption: 631.69 nm

Emission: 681.10 nm | PLQY: 0.330

ϵ : 13600 M⁻¹ cm⁻¹ | Reward: 1.003537